

Some Basic Physics, Chemistry and Geology followed by and Introduction to the Earth Interior



Today's Quiz

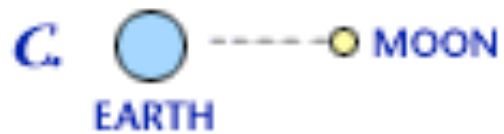
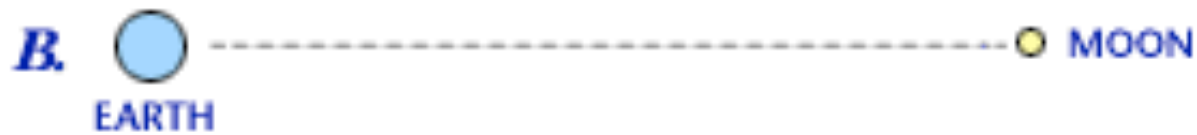
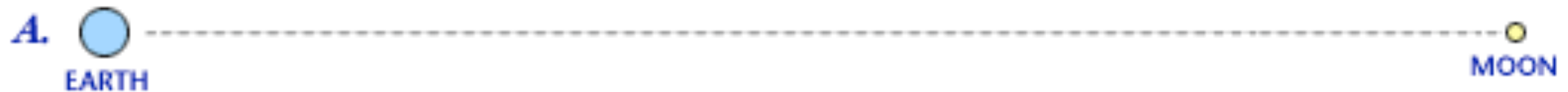
- 1) Vocabulary Chapters 1a and 11
- 2) Review of Chapter 1a and Previous Lecture



A Private Universe?

5 of 5

Which of the following diagrams most closely depicts the distance between Earth and the Moon?



- ☐ Diagram A
- ☐ Diagram B
- ☐ Diagram C



Some Basic Physics and Chemistry and Geology

Equilibrium

Cookie Equilibrium



Coffee Equilibrium



Equilibrium

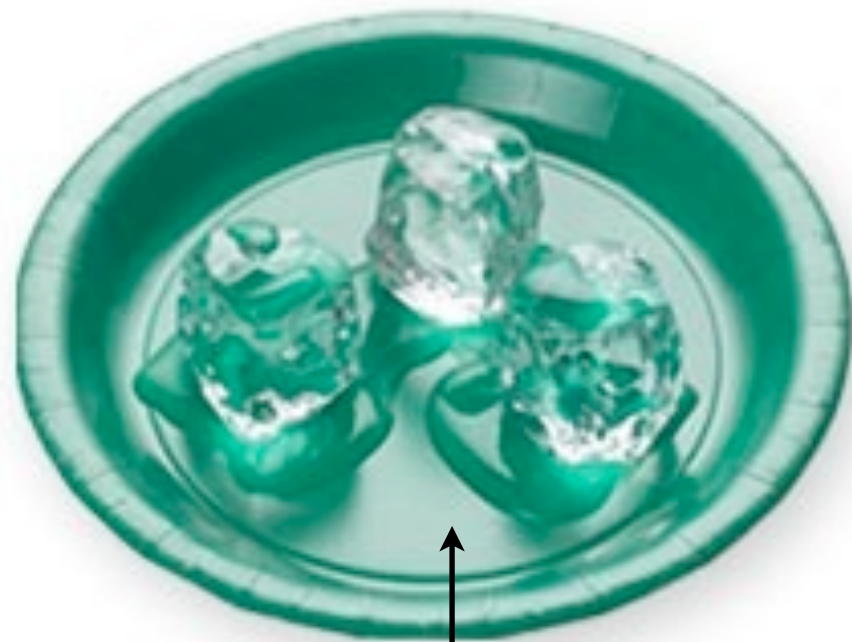


Molecular (H_2O) Equilibrium



Temperature (heat) Equilibrium

Phase Change



Solid



Gas

Liquid

Temperature (heat) and Pressure



Temperature is a measure of Heat



Accretionary heat
(energy transfer)

Sources of Earth's Heat

Accretionary heating (original heat)

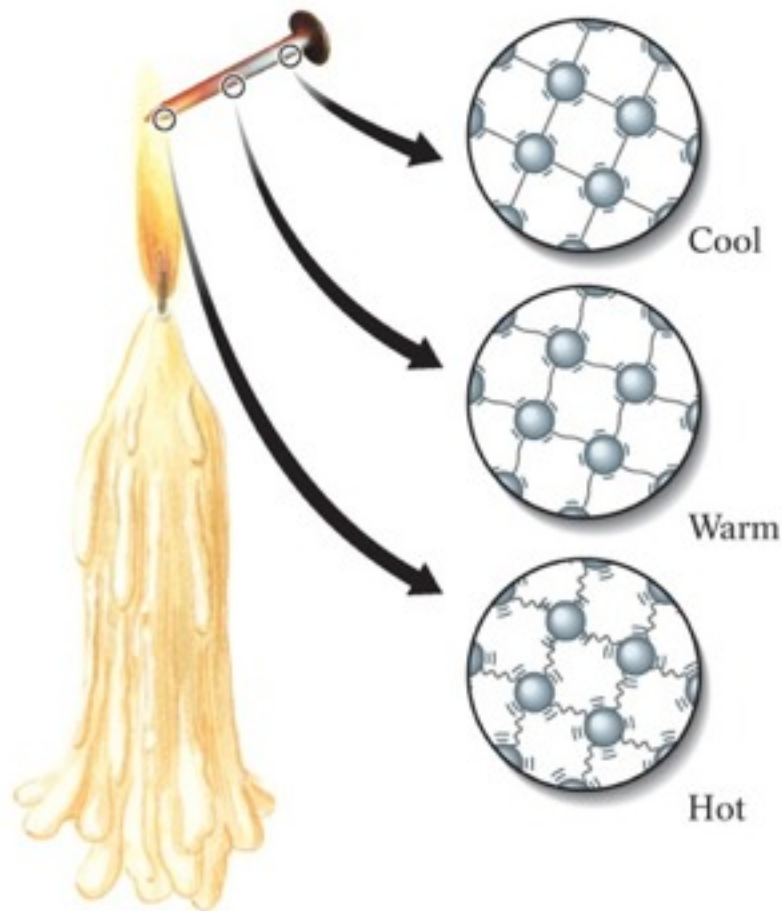
Latent heat (heat of crystallization)

Radioactive Decay



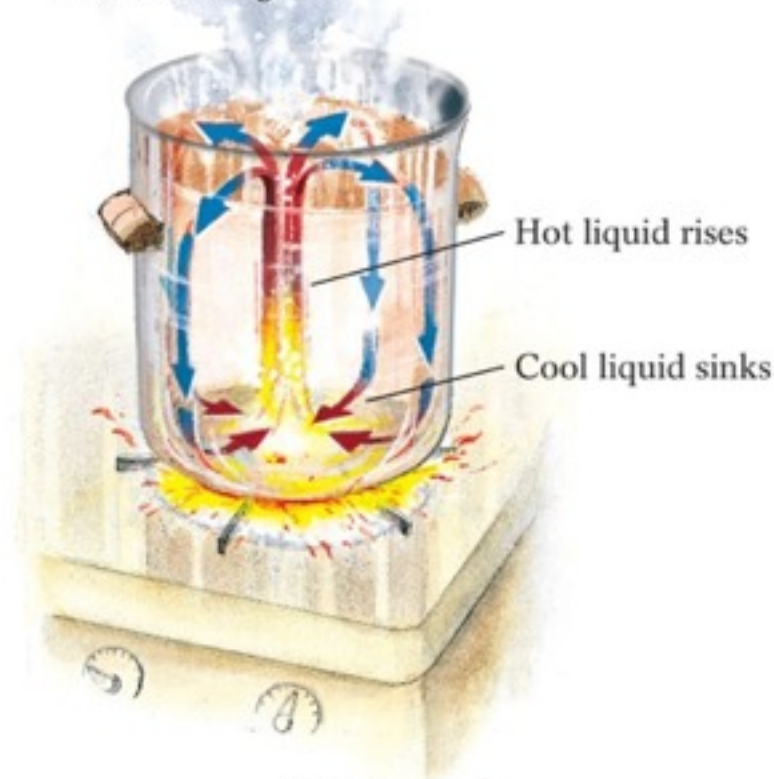
Heat Transfer (cooling)

Conduction of heat involves the passage of thermal energy from atom to neighboring atom. Thus, the atoms in the hot part of the nail, which is directly in the flame, vibrate very rapidly. Farther from the heat source, the atoms vibrate less rapidly. And at the end of the nail, which is still cool, the atoms vibrate very slightly. Eventually the heat will be conducted throughout the length of the nail.



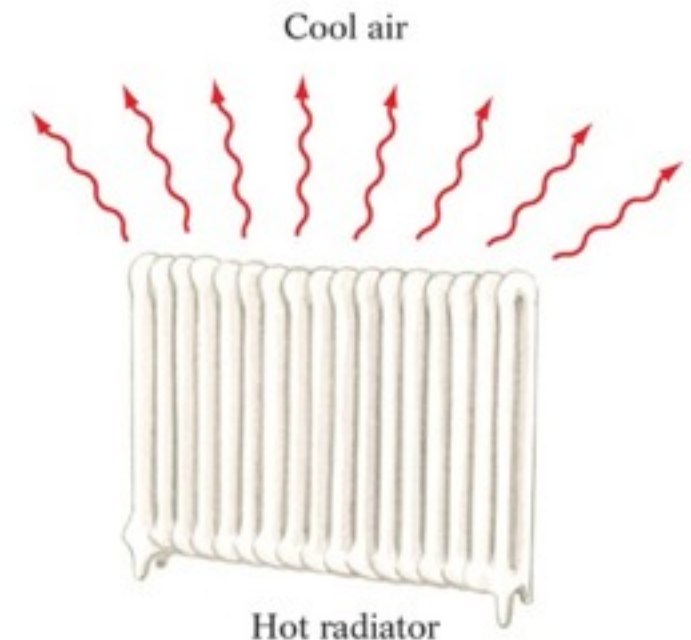
(a) Conduction

Convection involves the movement of heat from place to place by a flowing medium. Because the soup at the bottom of the pot is closer to the flame, it is the first part of the soup to become hot. As it heats up, it expands (becoming less dense) and rises, and the cooler (more dense) soup above it sinks to the bottom, displacing it and forcing it upward. When the warm soup arrives at the top, it encounters the relative coolness of the air and contracts in volume (becomes more dense) as it begins to cool. The cooled soup then sinks back toward the bottom, to be reheated and then to rise again.



(b) Convection

Radiation involves the transfer of heat from a hot object to its cooler surroundings. The hot radiator heats up the cool air that surrounds it.



(c) Radiation

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Buoyancy



Gravitational Equilibrium

$$\text{Density} = \frac{\text{mass}}{\text{volume}} \text{ (g/cm}^3\text{)}$$

9

Buoyancy



$$\text{Density} = \frac{\text{mass}}{\text{volume}} \text{ (g/cm}^3\text{)}$$

Buoyancy



Brass
on
Mercury

Gravitational Equilibrium

$$\text{Density} = \text{mass/volume (g/cm}^3\text{)}$$

Buoyancy



Brass
on
Mercury

$$\text{Density} = \text{mass/volume (g/cm}^3\text{)}$$

10

Observable Matter

Periodic Table of Elements															
1	2													3	4
1	2													3	4
3	4													5	6
11	12													13	14
19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34
37	38	39	40	41	42	43	44	45	46	47	48	49	50	51	52
55	56	57	72	73	74	75	76	77	78	79	80	81	82	83	84
87	88	89	104	105	106	107	108	109	110						

* Lanthanide Series

+ Actinide Series

58	59	60	61	62	63	64	65	66	67	68	69	70	71
Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu
90	91	92	93	94	95	96	97	98	99	100	101	102	103
Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr

Observable Matter

Periodic Table of Elements

1A	1	H	2	He	0																																
	3	Li	4	Be																																	
	5	B	6	C	7	N	8	O	9	F	10	Ne																									
	11	Na	12	Mg	IIIA	13	Al	14	Si	15	P	16	S	17	Cl	18	Ar																				
	19	K	20	Ca	IIIB	21	Sc	22	Ti	23	V	24	Cr	25	Mn	26	Fe	27	Co	28	Ni	29	Cu	30	Zn	31	Ga	32	Ge	33	As	34	Se	35	Br	36	Kr
	37	Rb	38	Sr	39	Y	40	Zr	41	Nb	42	Mo	43	Tc	44	Ru	45	Rh	46	Pd	47	Ag	48	Cd	49	In	50	Sn	51	Sb	52	Te	53	I	54	Xe	
	55	Cs	56	Ba	*La	72	Hf	73	Ta	74	W	75	Re	76	Os	77	Ir	78	Pt	79	Au	80	Hg	81	Tl	82	Pb	83	Bi	84	Po	85	At	86	Rn		
	87	Fr	88	Ra	+Ac	104	Rf	105	Ha	106	107	108	109	110																							

* Lanthanide Series

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58	59	60	61	62	63	64	65	66	67	68	69	70	71
Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu
90	91	92	93	94	95	96	97	98	99	100	101	102	103
Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr

Silicate Rocks- Rocks composed dominantly of Silicon (Si) and Oxygen (O)

Observable Matter

Felsic rocks

Periodic Table of Elements

Mafic rocks

1A	1 H	2 He	0																																
1A	3 Li	4 Be	5A	6A	7A	8A	9A	10A	11A	12A	13A	14A	15A	16A	17A	18A																			
1A	11 Na	12 Mg	13 Al	14 Si	15 P	16 S	17 Cl	18 Ar	19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr									
1A	37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe	55 Cs	56 Ba	57 *La	58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu
1A	87 Fr	88 Ra	89 +Ac	90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr	104 Rf	105 Ha	106 106	107 107	108 108	109 109	110 110	111 111	112 112	113 113	114 114	115 115	116 116	117 117	118 118			

* Lanthanide Series

+ Actinide Series

58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu
90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr

Silicate Rocks- Rocks composed dominantly of Silicon (Si) and Oxygen (O)

Observable Matter

Earth's Composition by Mass

- 35% Fe
- 30% O
- 15% Si
- 12% Mg
- 8% all other

Felsic rocks

Periodic Table of Elements

Mafic rocks

1A	1 H	2A																0 2 He	
1	3 Li	4 Be										5 B	6 C	7 N	8 O	9 F	10 Ne		
2	11 Na	12 Mg	III B	IV B	V B	VIB	VII B	VIII	IX	X	IB	13 Al	14 Si	15 P	16 S	17 Cl	18 Ar		
3	19 K	20 Ca		21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr
4	37 Rb	38 Sr		39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe
5	55 Cs	56 Ba	*La	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn	
6	87 Fr	88 Ra	+Ac	104 Rf	105 Ha	106	107	108	109	110									
7																			

* Lanthanide Series

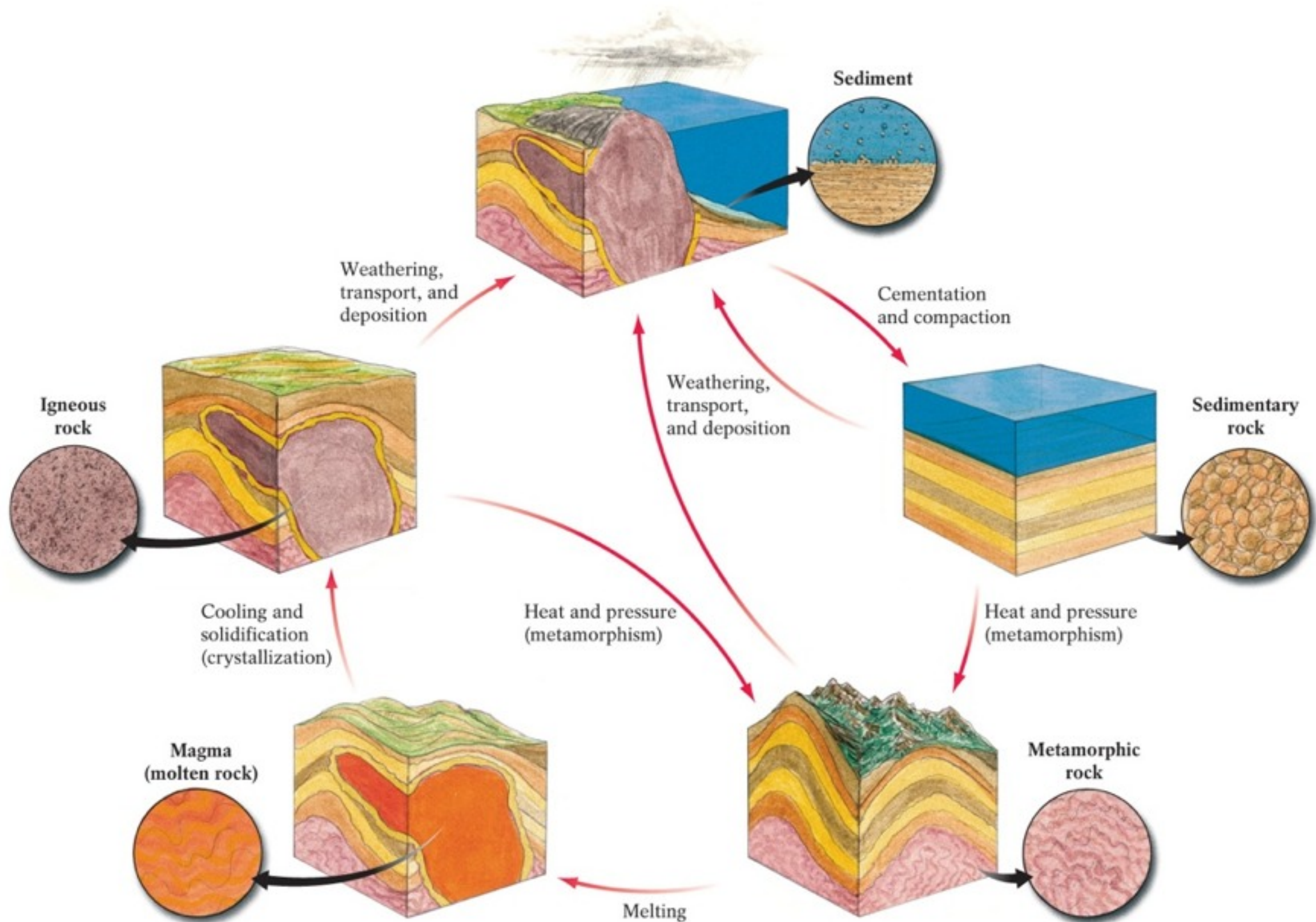
58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu
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+ Actinide Series

90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr
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Silicate Rocks- Rocks composed dominantly of Silicon (Si) and Oxygen (O)

The Rock Cycle



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Observable Matter

Earth's Composition by Mass

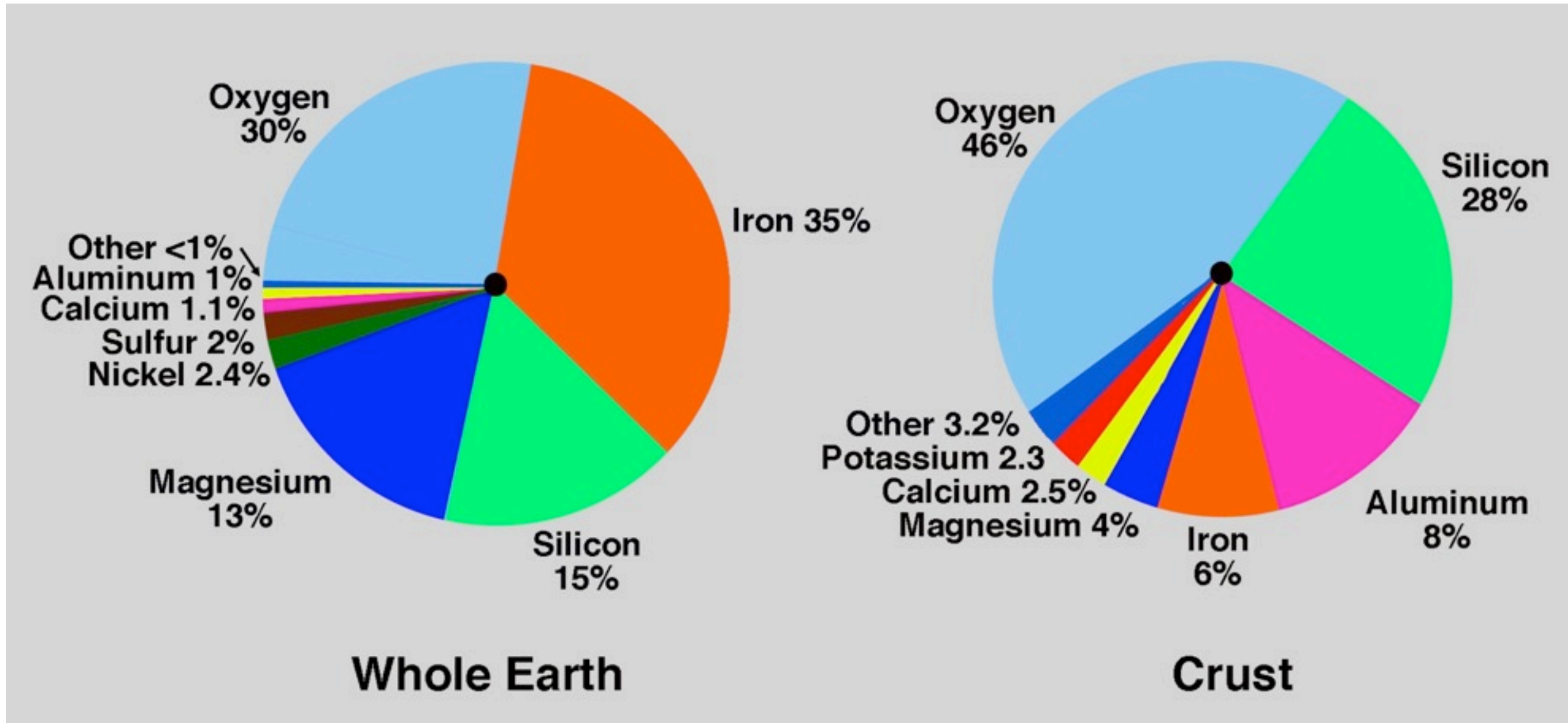
- 35% Fe
- 30% O
- 15% Si
- 12% Mg
- 8% all other

The image shows a standard periodic table of elements. Two blue ovals highlight specific groups of elements. The first oval, labeled 'Felsic rocks' in blue text, encompasses the elements Boron (B), Carbon (C), Nitrogen (N), Oxygen (O), Silicon (Si), and Phosphorus (P). The second oval, labeled 'Mafic rocks' in blue text, encompasses the elements Sodium (Na), Magnesium (Mg), Calcium (Ca), and Iron (Fe). The elements Oxygen (O), Silicon (Si), and Iron (Fe) are each enclosed in a red square. The periodic table includes element symbols, atomic numbers, and group labels (IA, IIA, IIIA, IVA, VA, VIA, VIIA, 0, IB, IIB, IIIB, IVB, VB, VIB, VIIB, VII, IB, IIB).

Felsic Rocks Greater than 55% SiO₂ (silica) → **Continental Crust**
 45% SiO₂ < **Mafic Rocks** < 55% SiO₂ → **Oceanic Crust**
Ultramafic less than 45% SiO₂ → **Mantle**
Non-Silicate Rock (Ni-Fe alloy) → **Core**

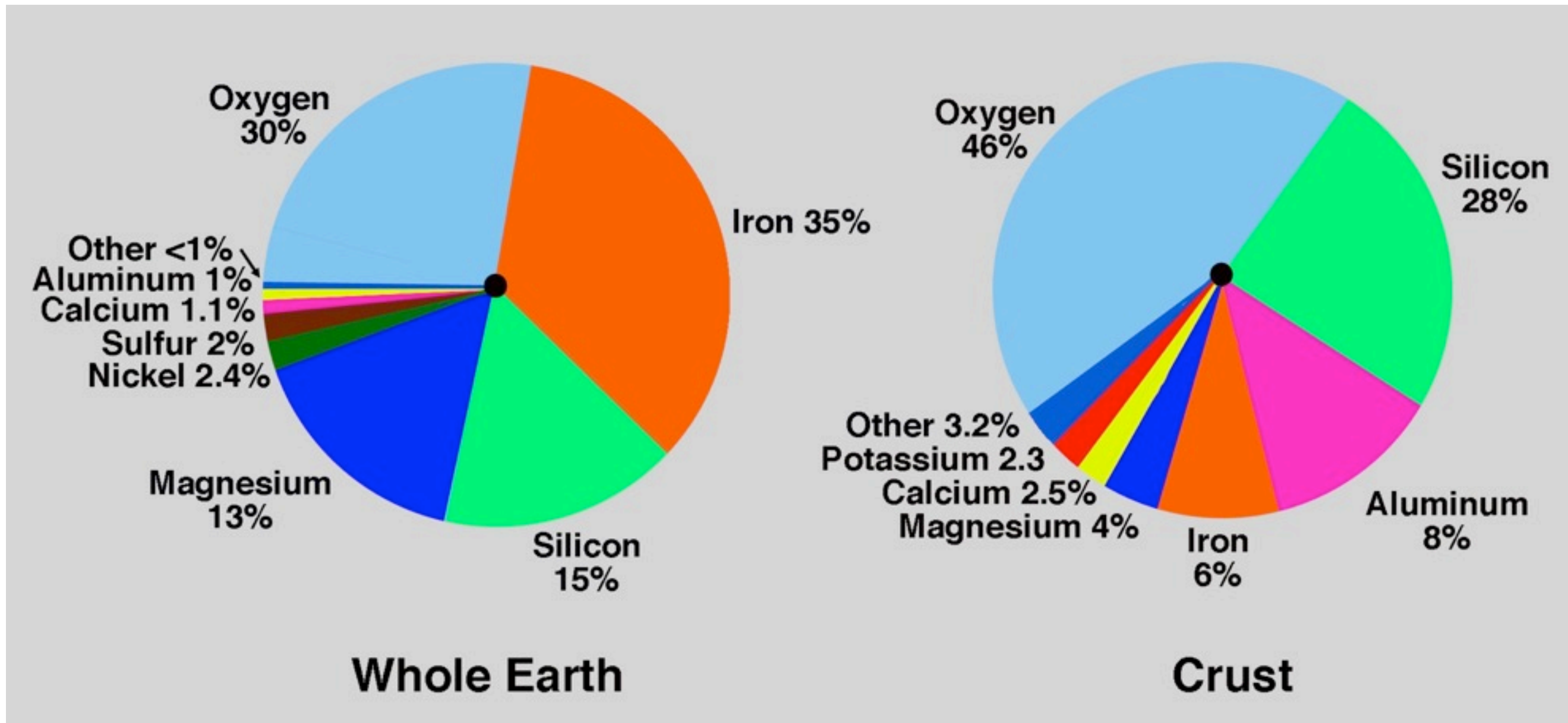
Earth's Composition

Heterogeneous or Homogeneous?



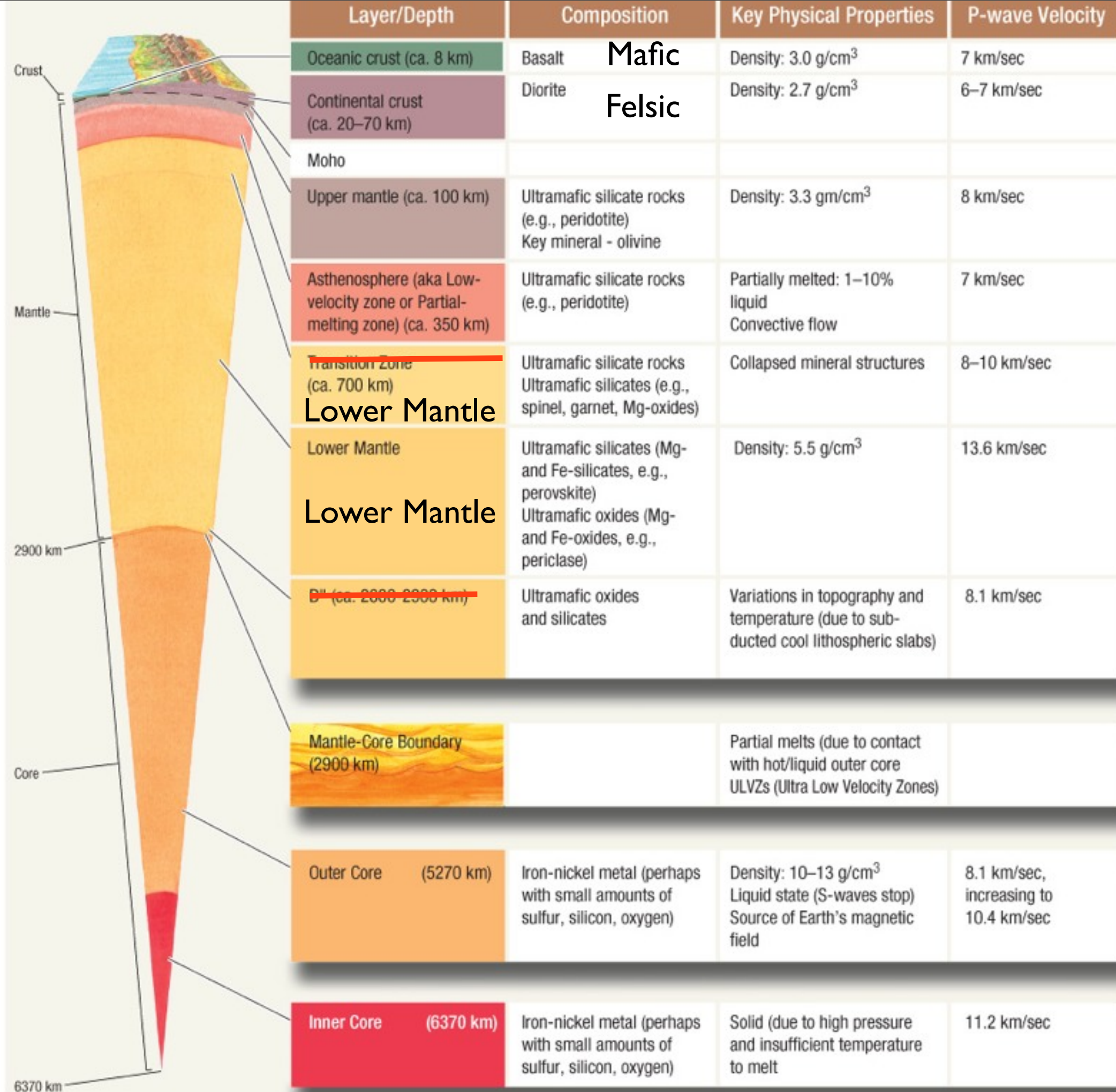
Earth's Composition

Heterogeneous or Homogeneous?

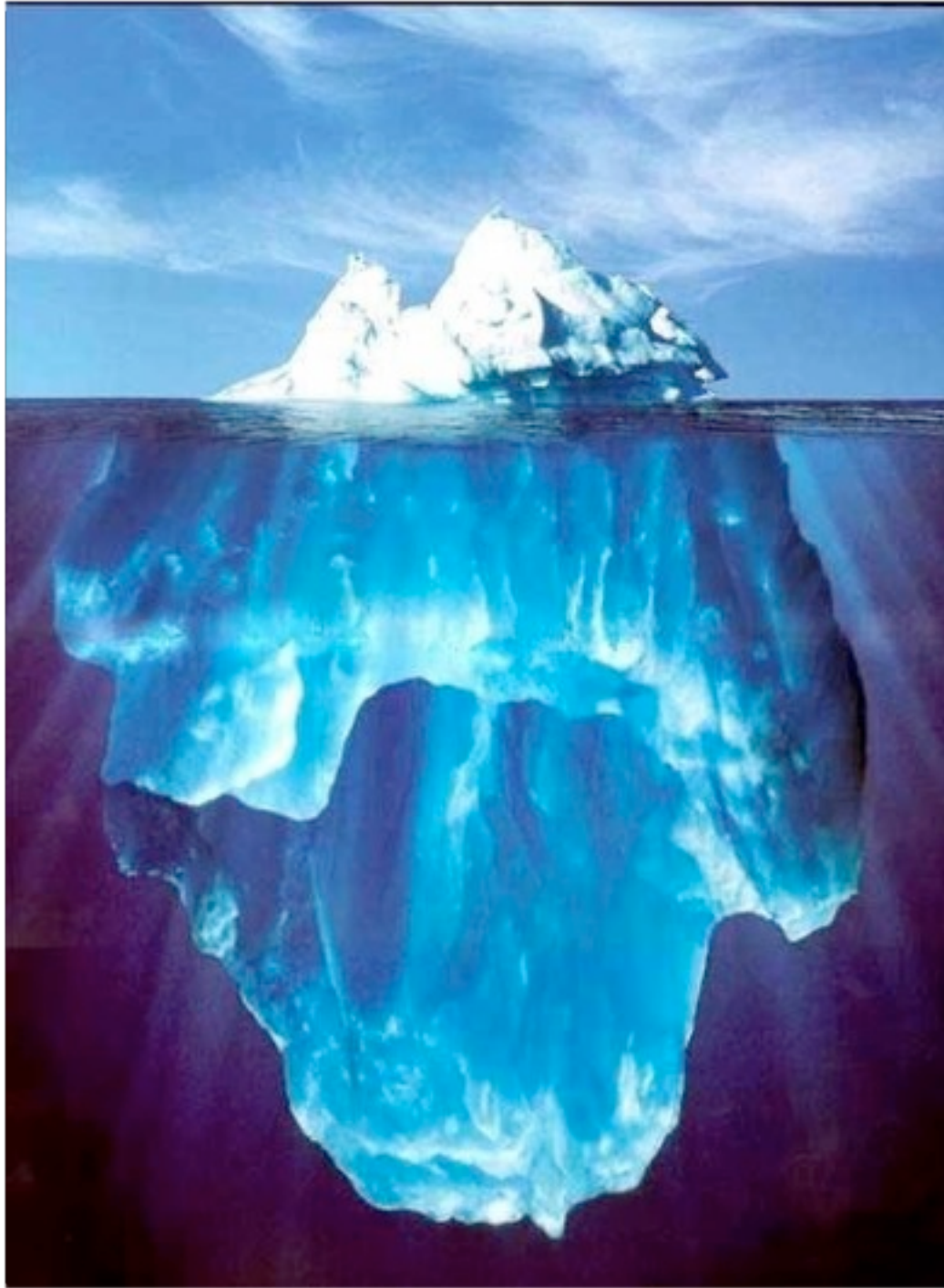


What is the crust enriched in?
What is the crust depleted in?

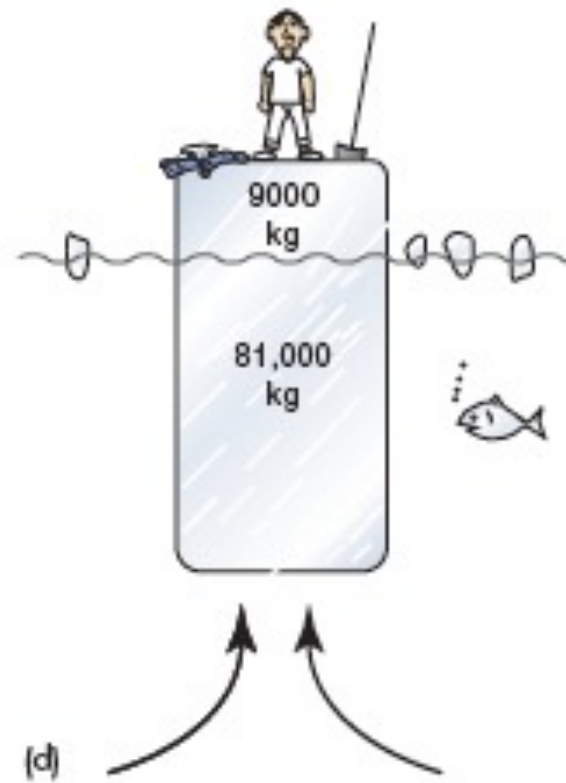
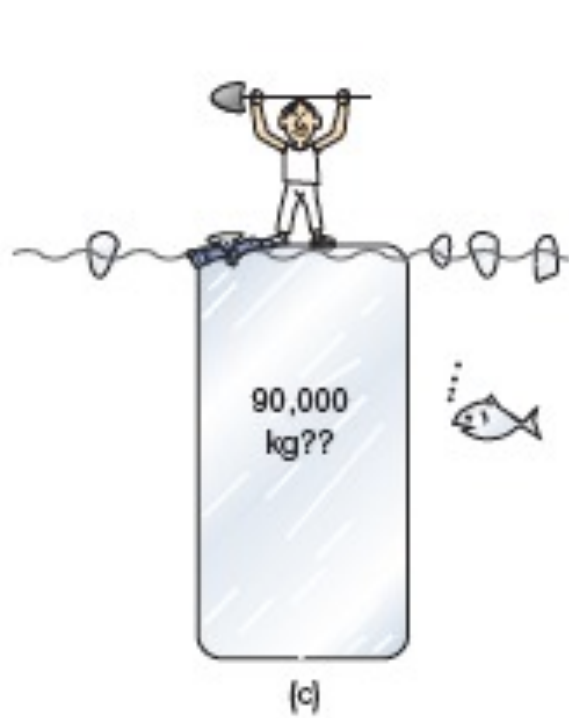
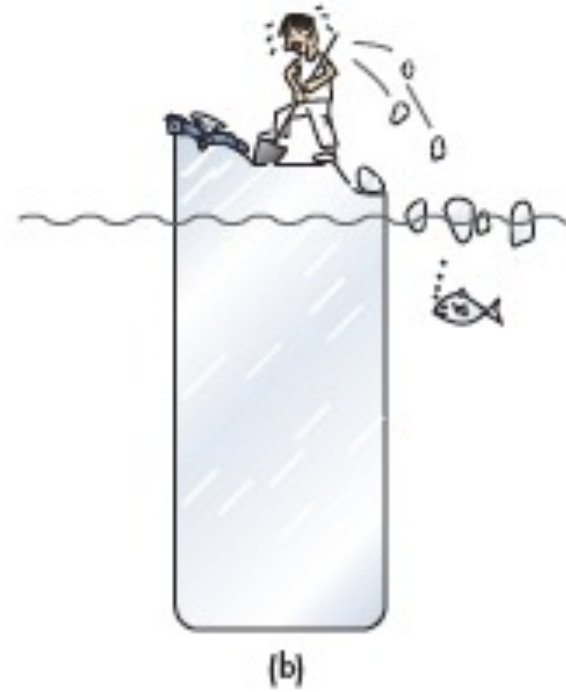
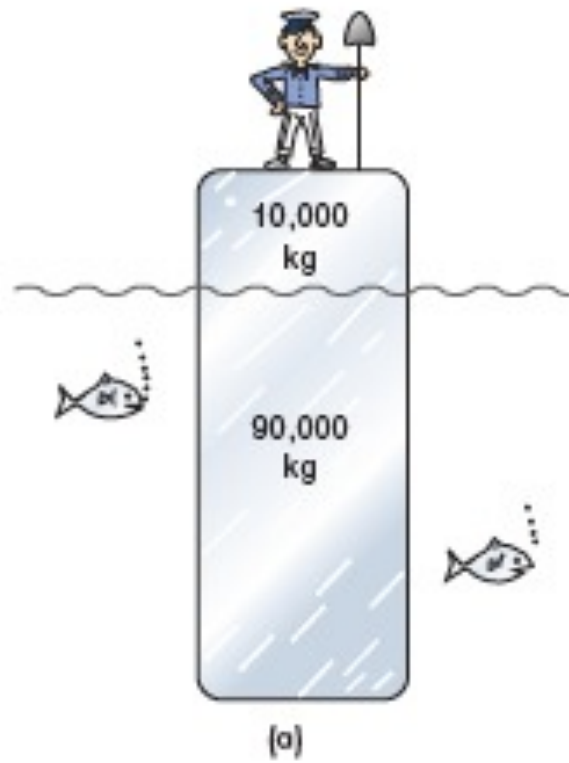
Modified “Geology at a Glance” I I.I



Buoyancy

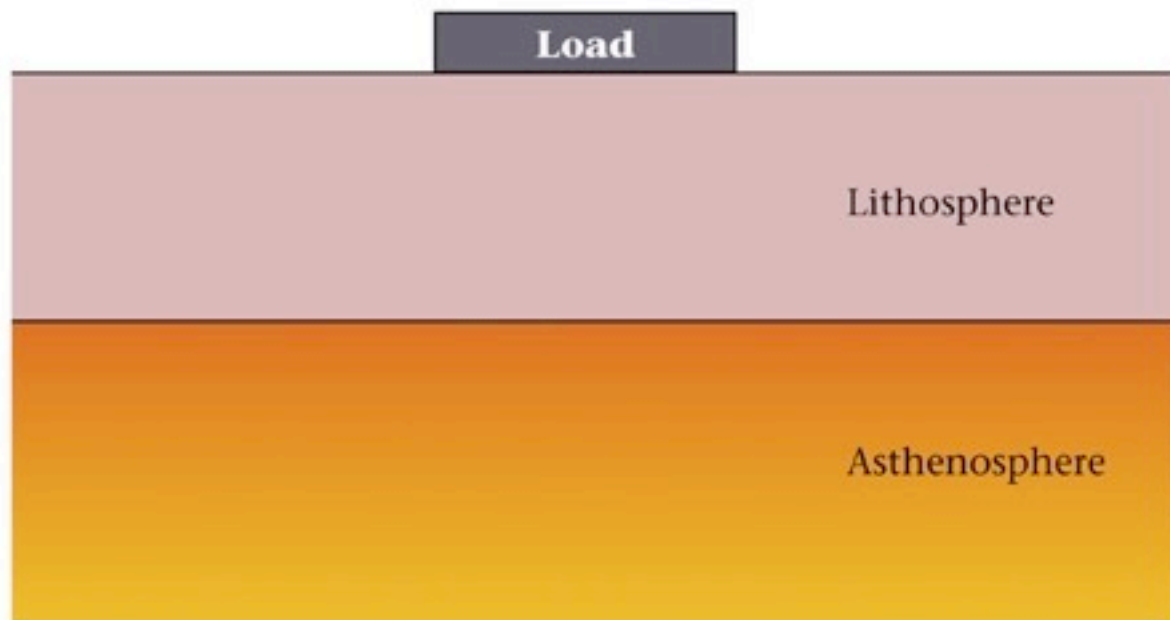


Isostasy

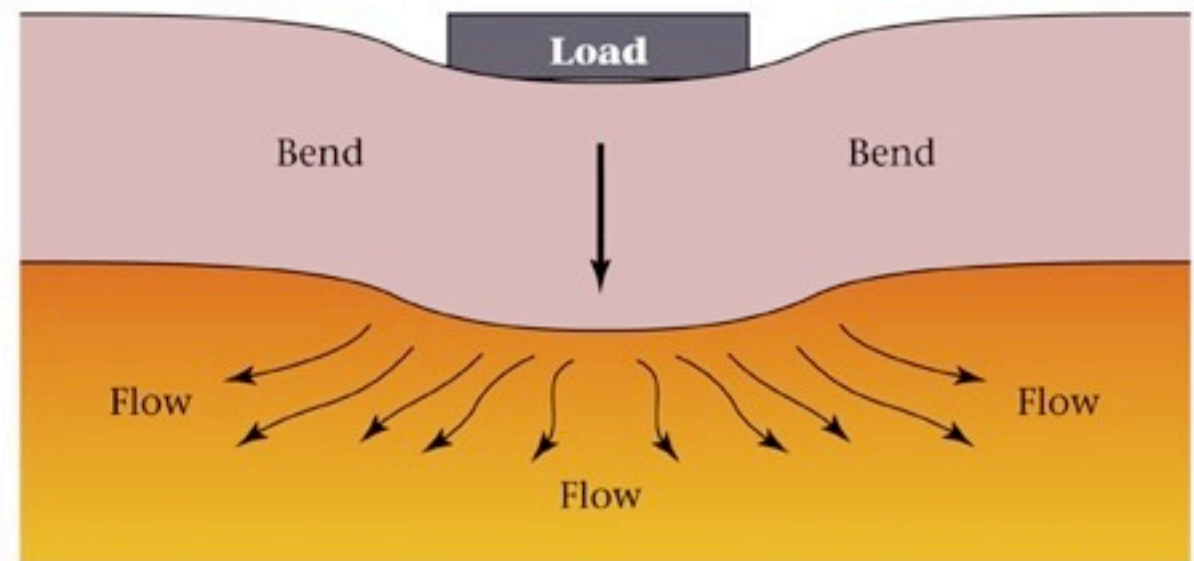


Isostasy

Time 1: A "load" is placed on top of the lithosphere.

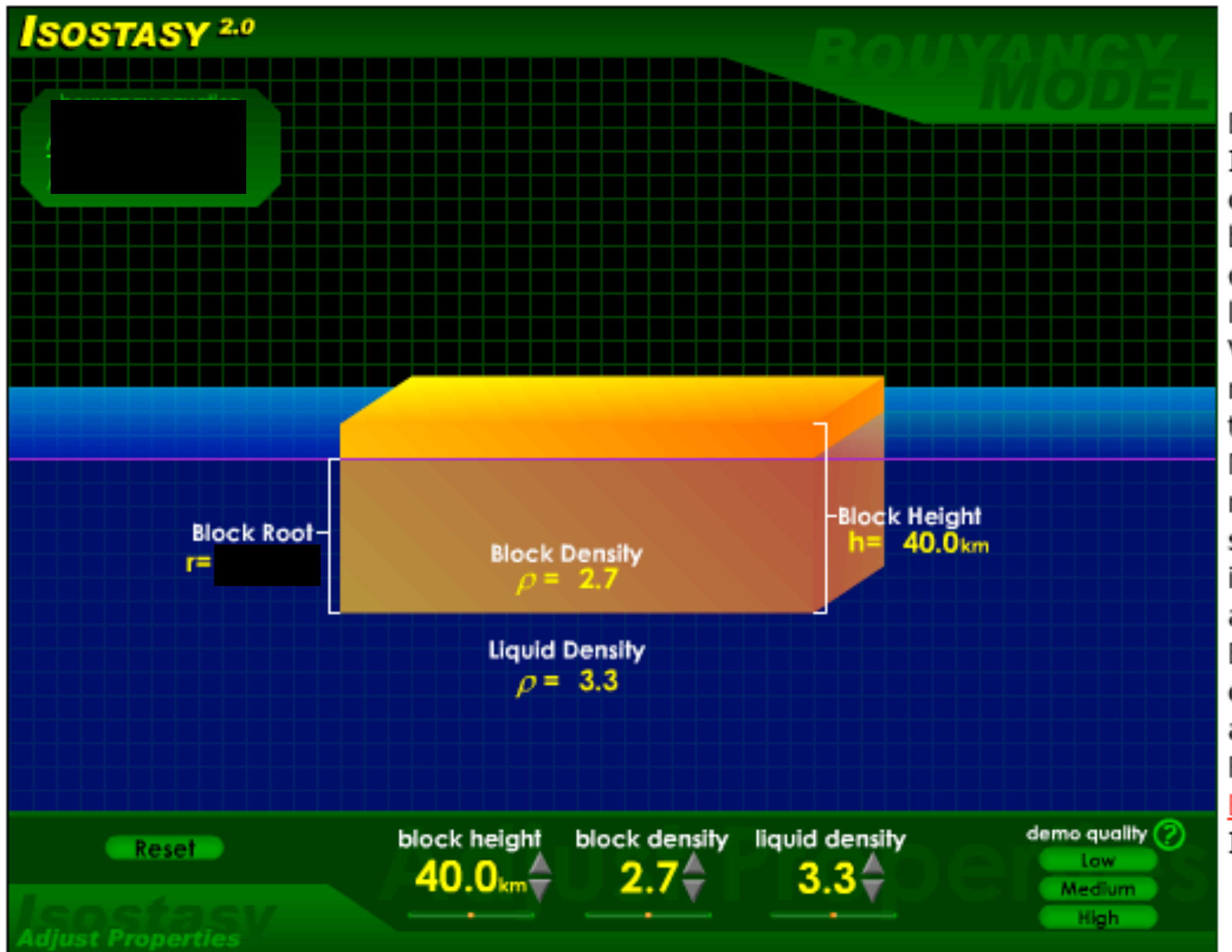


Time 2: The weight of the load pushes down. The lithosphere bends and its base moves down. The plastic asthenosphere flows out of the way.



(not to scale)

Isostasy



What amount of the low density block is sticking above the more dense material?

a) ~12 km

b) ~33 km

c) ~17 km

d) ~7 km

e) Was unable to determine

What amount of the low density block is sticking above the more dense material?

a) ~12 km

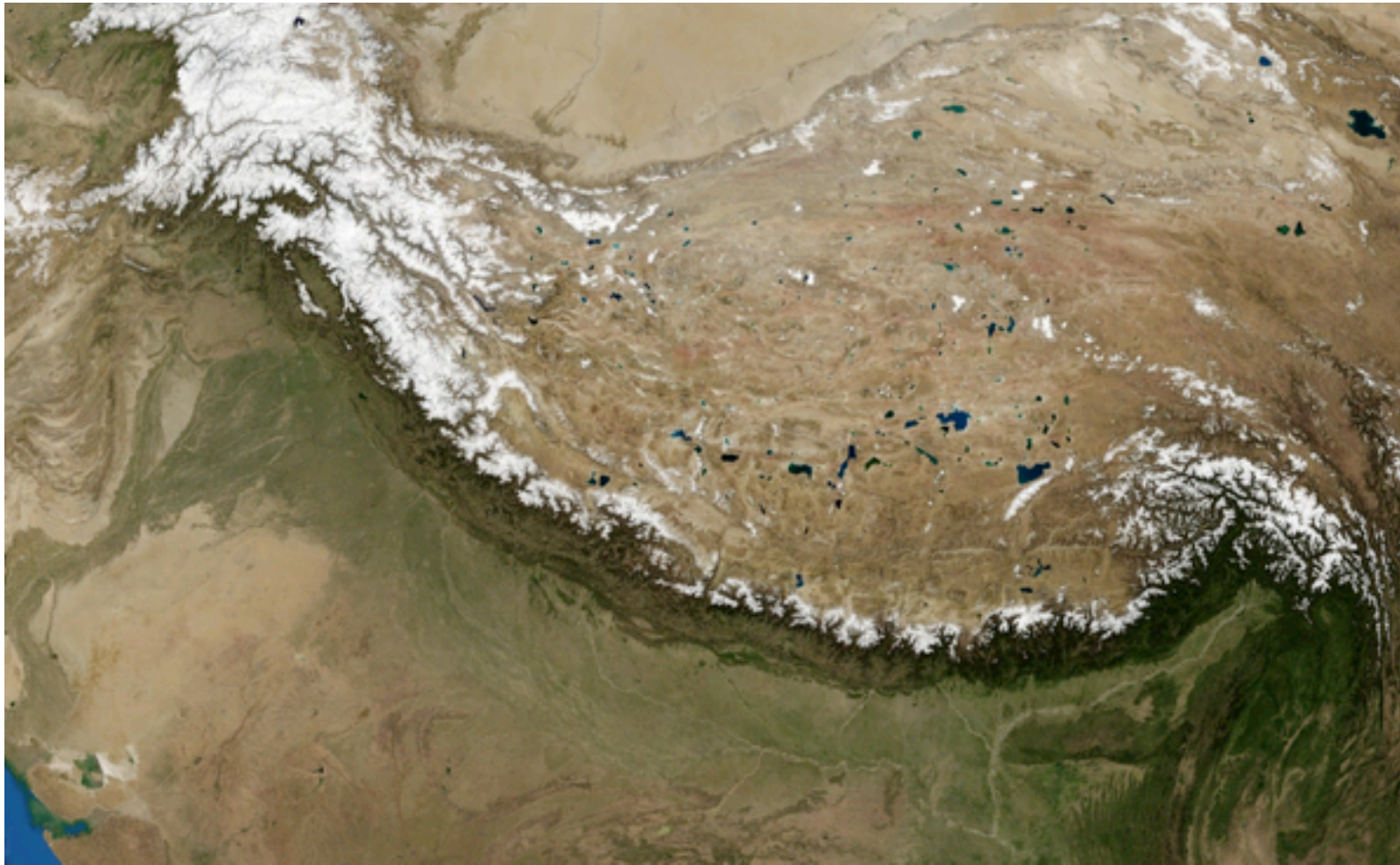
b) ~33 km

c) ~17 km

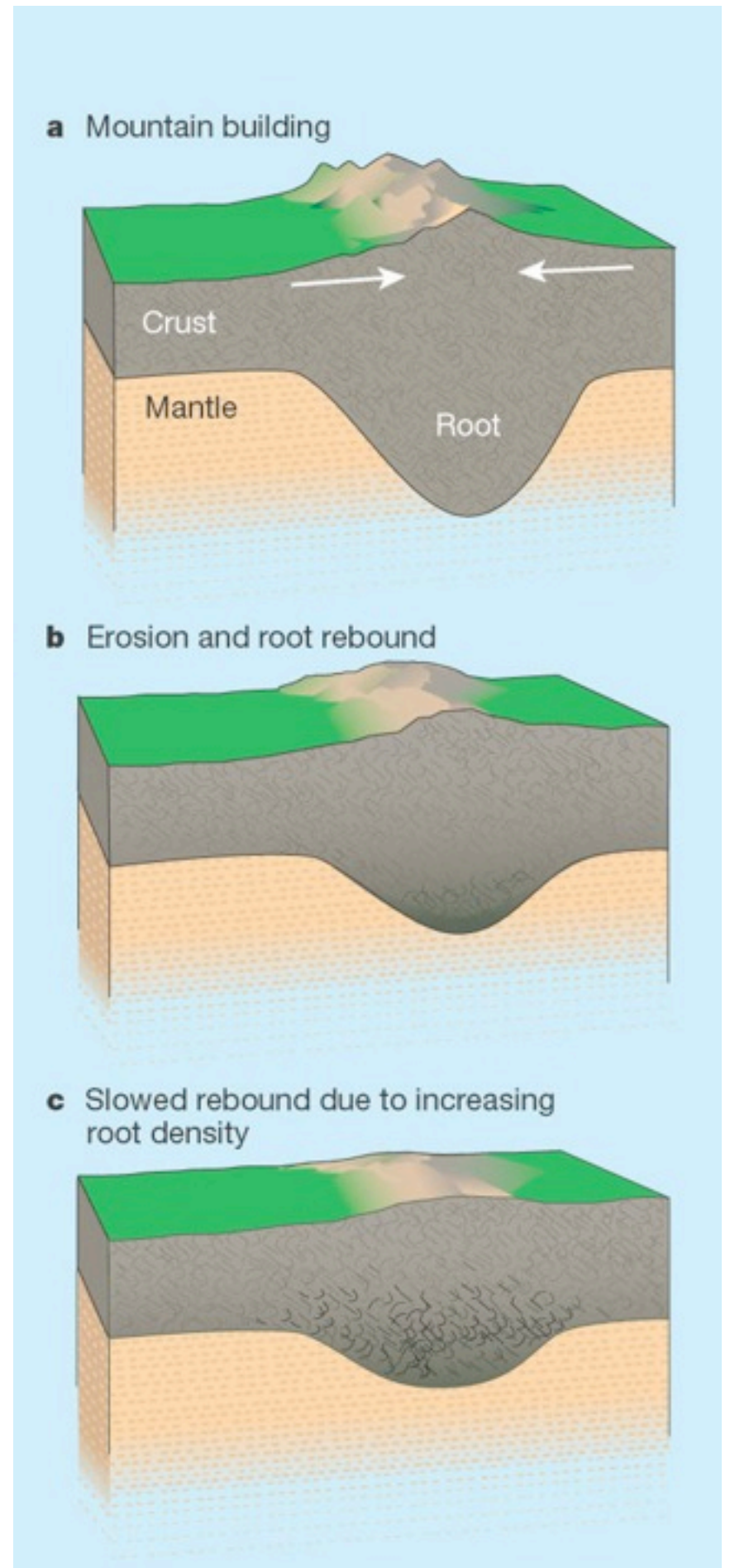
d) ~7 km

e) Was unable to determine

Isostasy



Himalayan Mountains



Earth's Structure Group Work

Pages 17, and 19-21 in your Manual
Work in Groups of 2 to 3

Earth's Structure

