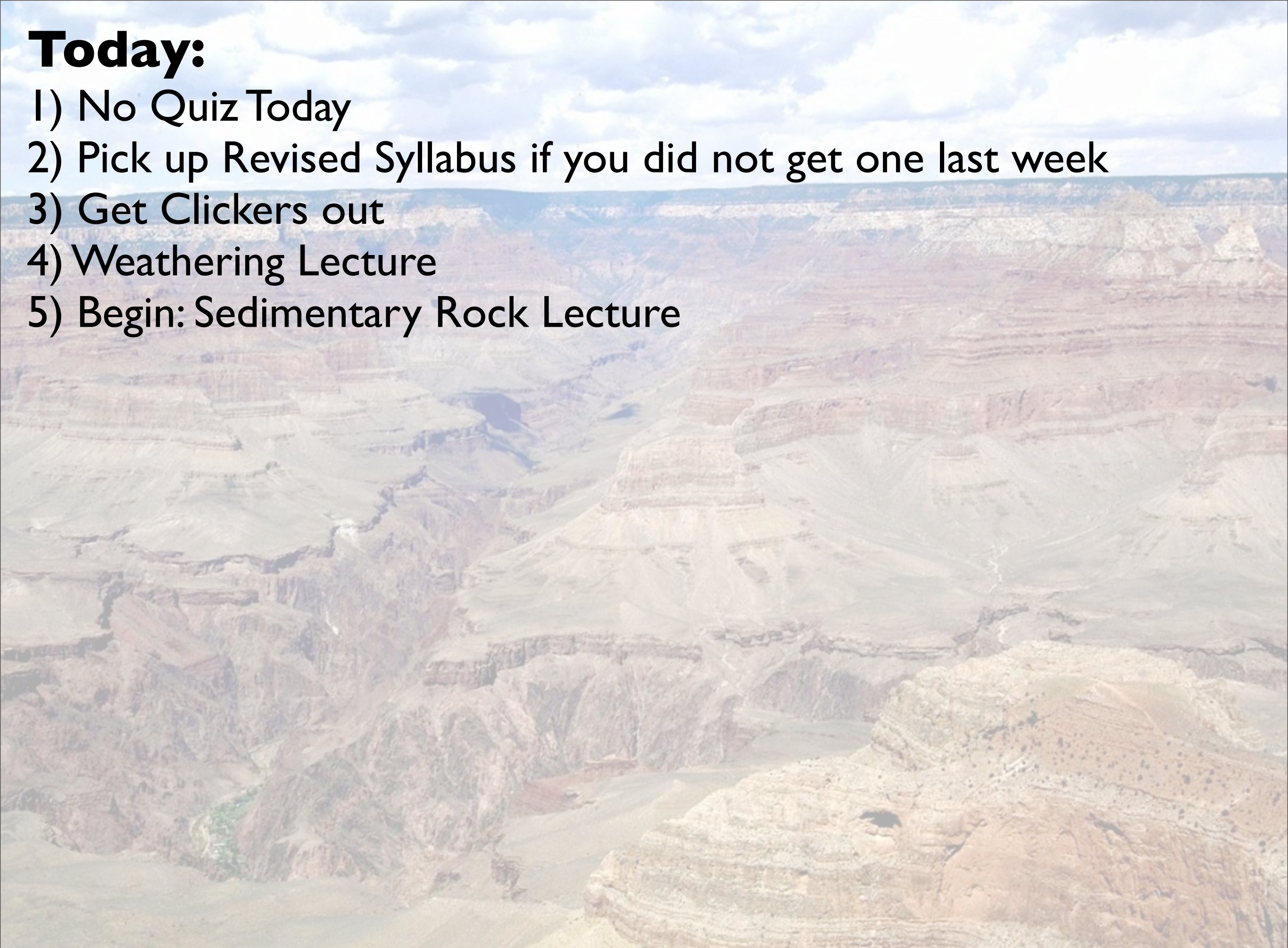


Today:

- 1) No Quiz Today
- 2) Pick up Revised Syllabus if you did not get one last week
- 3) Get Clickers out
- 4) Weathering Lecture
- 5) Begin: Sedimentary Rock Lecture



Just from looking at the shape of this Volcano, which is the best description?

A) low viscosity, high volatiles, high explosivity

B) high viscosity, high volatiles, low explosivity

C) high viscosity, low volatiles, high explosivity

D) high viscosity, high volatile, high explosivity

E) none of the above



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B) high viscosity, high volatiles, low explosivity

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E) none of the above



What type of volcano is this?

- A) Volcanic dome
- B) Continental caldera
- C) Scoria cone
- D) Shield volcano
- E) Stratovolcano



What type of volcano is this?

- A) Volcanic dome
- B) Continental caldera
- C) Scoria cone
- D) Shield volcano
- E) Stratovolcano



This volcano is probably located at_____.

A) convergent continental collision zone

B) continental rift

C) divergent boundary

D) convergent subduction zone

E) none of the above



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The high viscosity of the lava that composes this volcano is probably the result of_____.

A) the location of where the melt was generated

B) partial melting of the asthenosphere

C) decompression of the asthenosphere

D) the high mafic content of the magma.

E) fractional crystallization



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E) fractional crystallization



A photograph of a river flowing through a canyon. The river is in the foreground, curving to the right. The canyon walls are composed of layered sedimentary rock, showing various shades of brown, orange, and grey. The text is overlaid on the upper half of the image.

Weathering, Erosion, and Deposition = Sedimentary Rocks A Story of Rock Recycling

Weathering

The process that breaks up and corrodes solid rock, in place, eventually transforming it into sediment.

Mechanical Weathering

reduction of rock/grain size by physically breaking (disintegration) without changing the composition.



Chemical weathering

conversion of unstable mineral and bulk rock composition (dissolving) into stable compounds at the Earth's surface.



(a)

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(b)

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Weathering

The process that breaks up and corrodes solid rock, in place, eventually transforming it into sediment.

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(a)

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(b)

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Physical/Mechanical Weathering

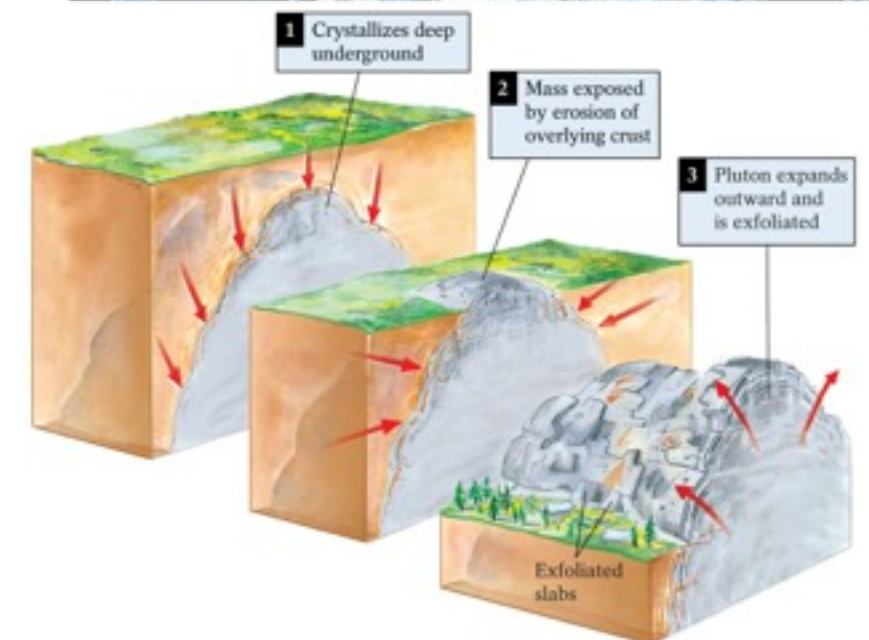
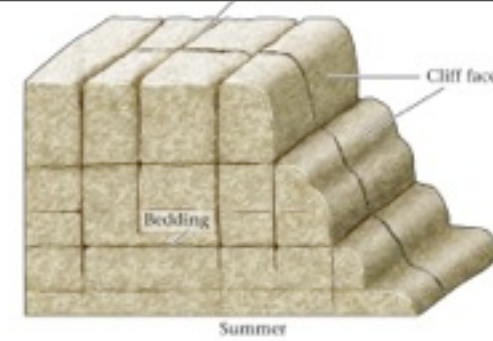
Frost Wedging

Root Wedging

Salt Wedging

Thermal expansion/contraction

Exfoliation



Physical/Mechanical Weathering

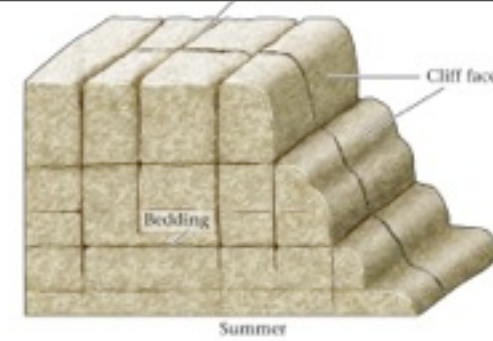
Frost Wedging

Root Wedging

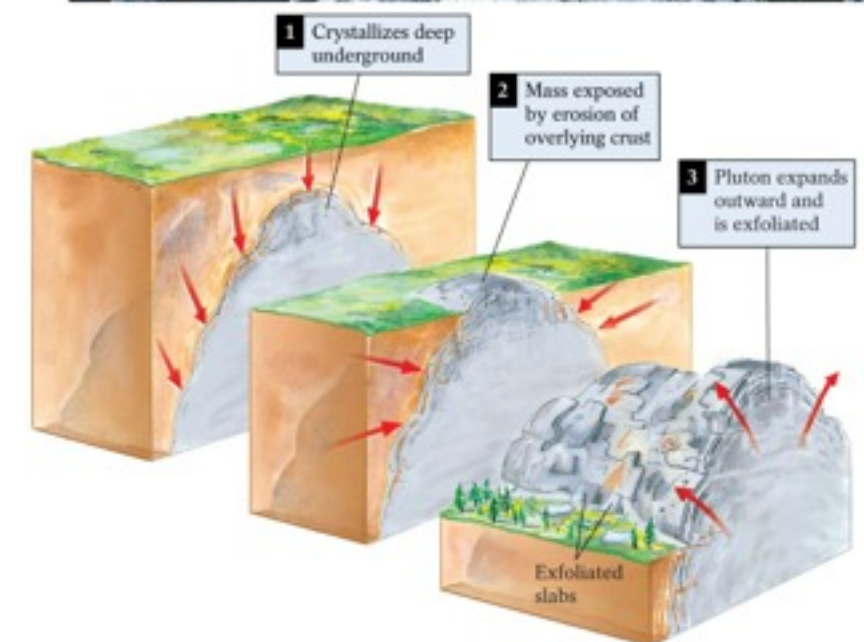
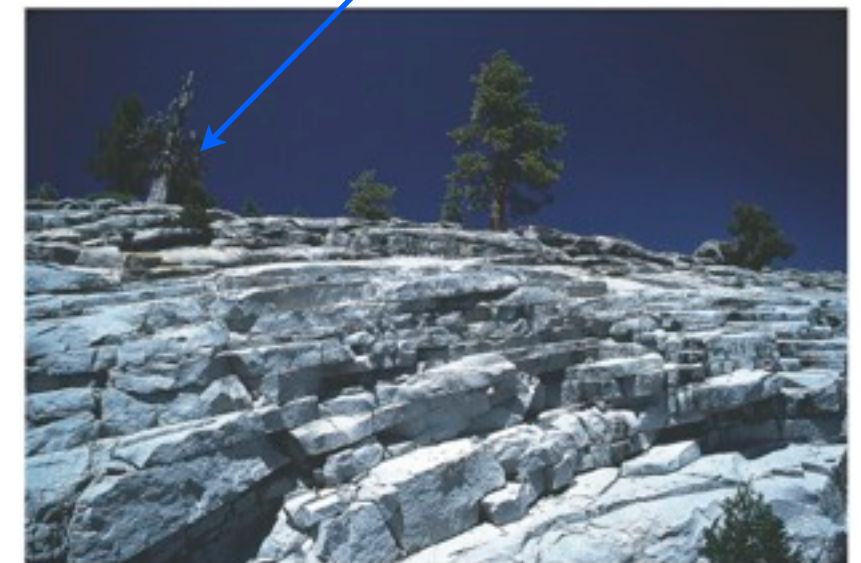
Salt Wedging

Thermal expansion/contraction

Exfoliation



Frost and root wedging takes advantage of exfoliation cracks



Physical/Mechanical Weathering

Frost Wedging

Root Wedging

Salt Wedging

Thermal expansion/contraction

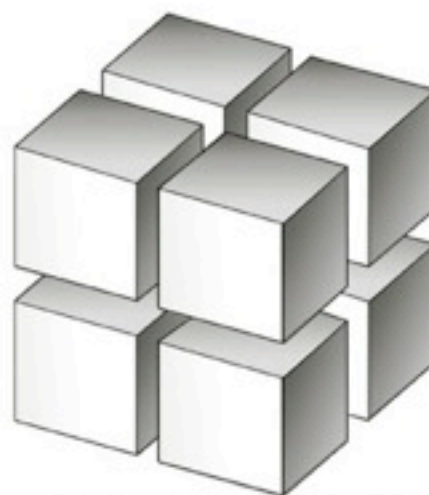
Exfoliation



Fewer cracks,
less surface area

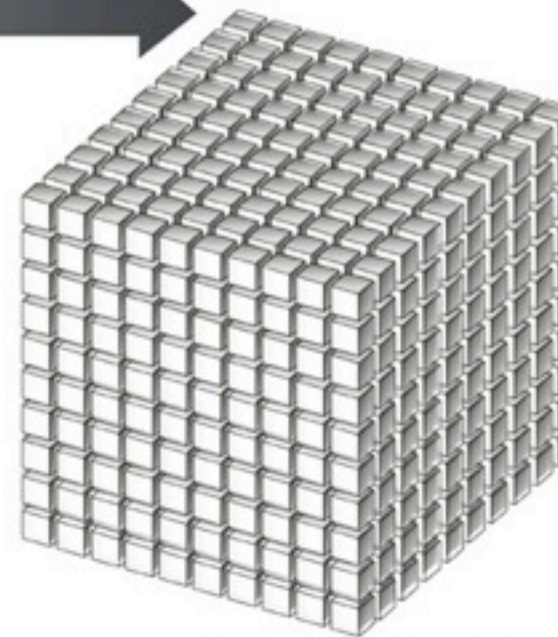


Surface area = 6 m^2



Surface area = 12 m^2

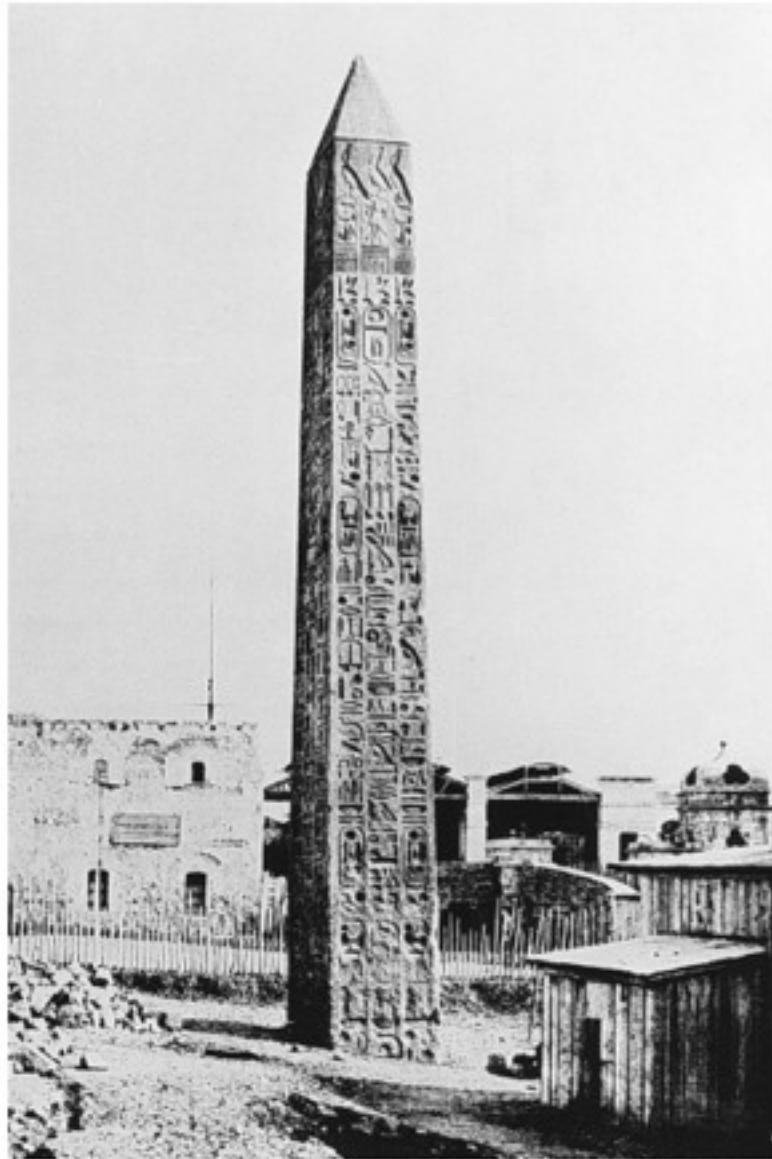
More cracks,
more surface area



Surface area = 60 m^2

Chemical Weathering

Dissolution
Hydrolysis
Oxidation



(a)



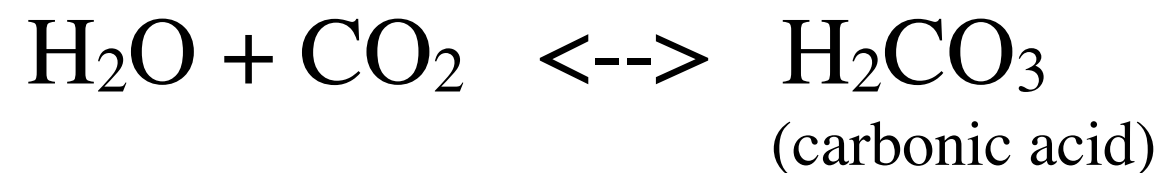
(b)

15



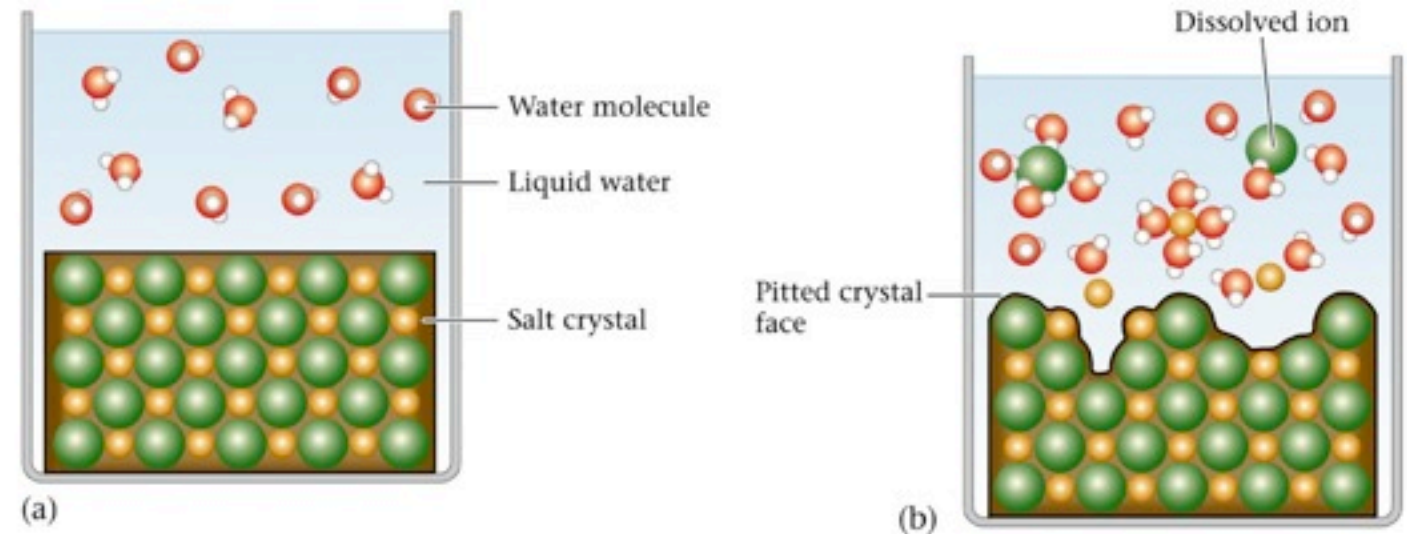
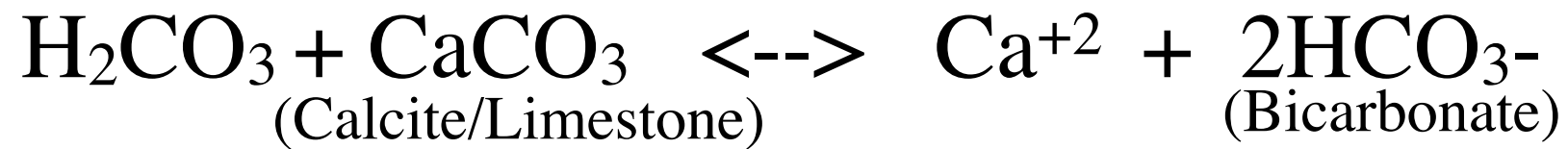
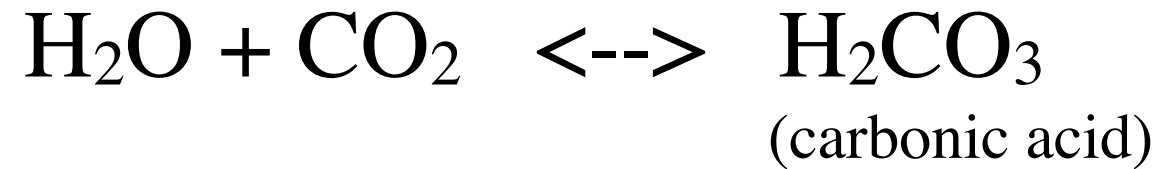
Chemical Weathering

Dissolution

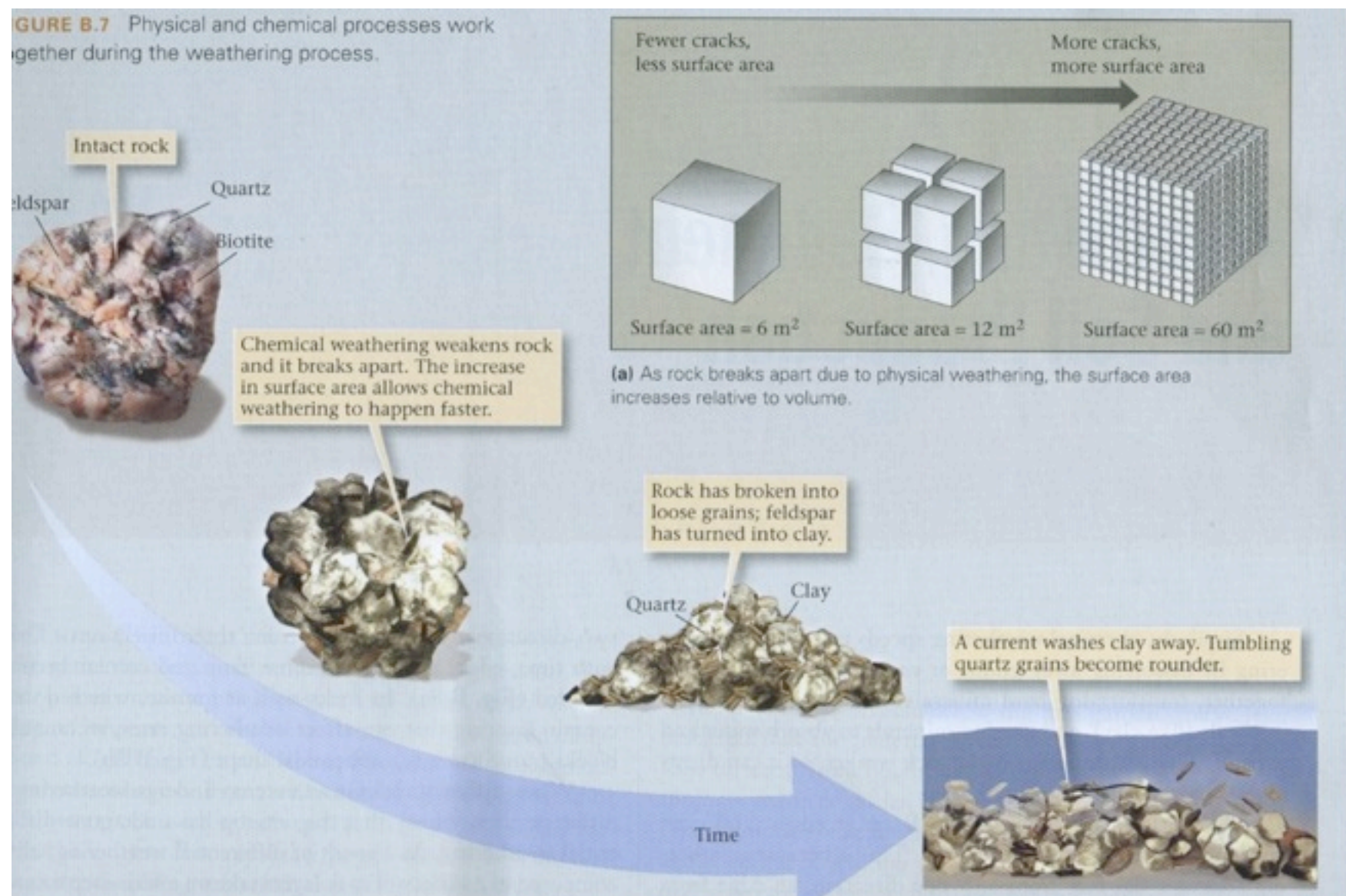


Chemical Weathering

Dissolution

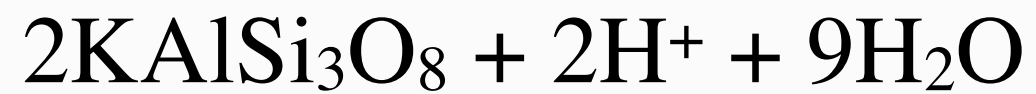


Hydrolysis



Chemical Weathering

Hydrolysis



Potassium

Feldspar



Chemical Weathering

Hydrolysis



Chemical Weathering

Oxidation

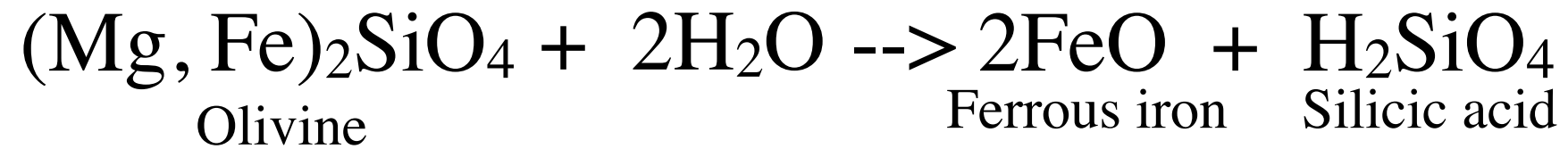


Olivine



Chemical Weathering

Oxidation

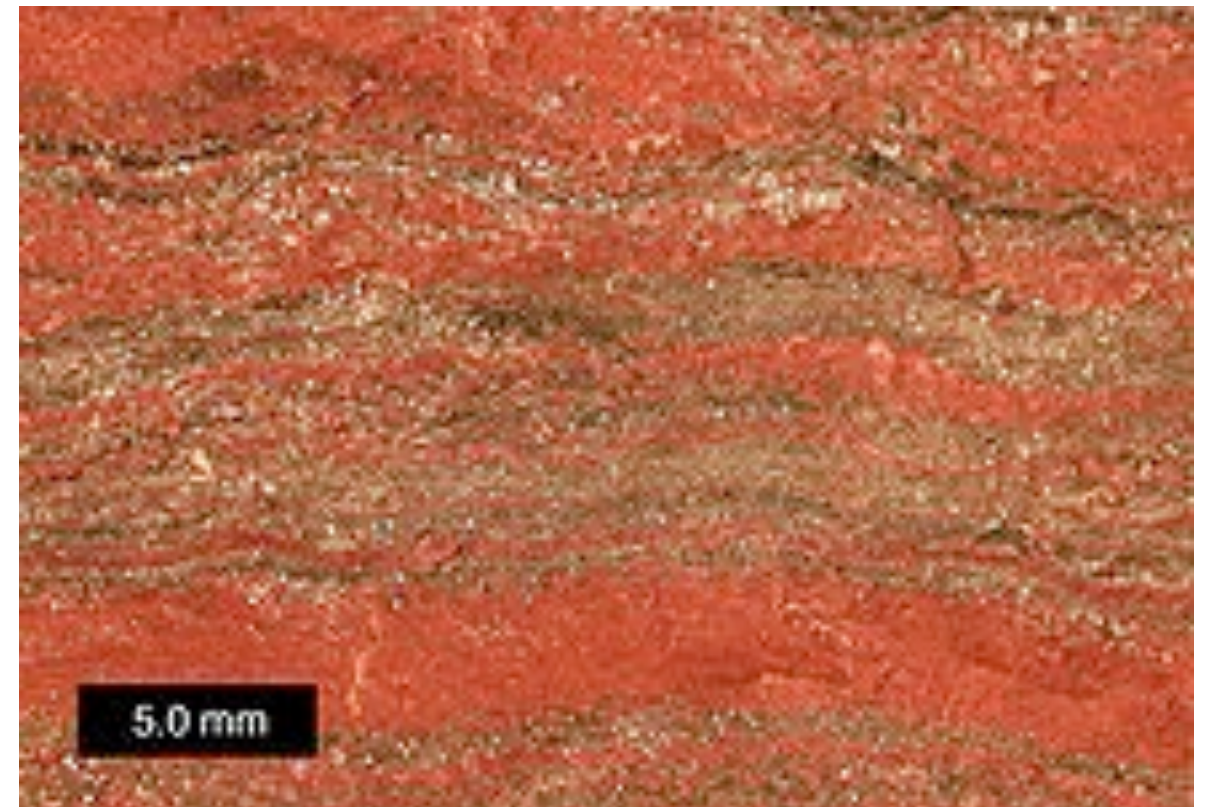
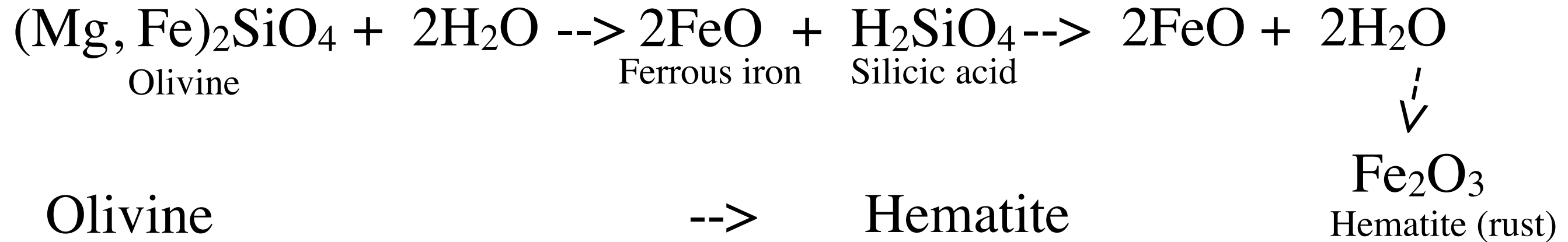


Olivine

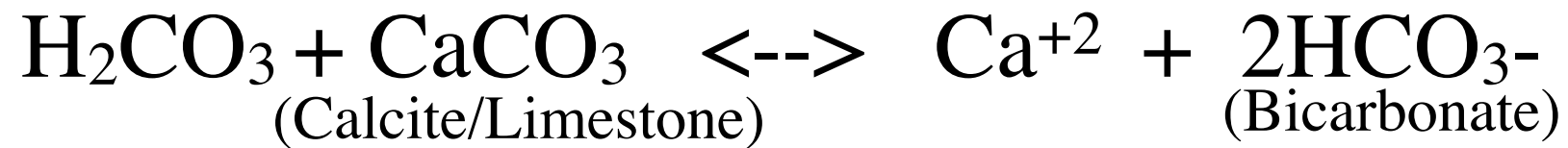


Chemical Weathering

Oxidation

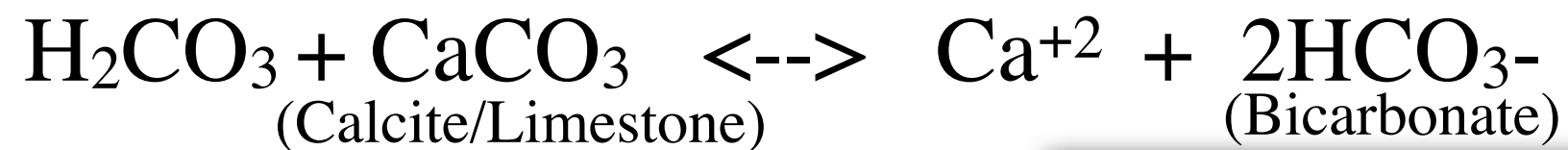


Where do the Ions go?



COMPARISON BETWEEN OCEAN WATER AND RIVER WATER		
Chemical Constituent	Percentage of Total Salt Content	
	Ocean Water	River Water
Silica (SiO_2)	—	14.51
Iron (Fe)	—	0.74
Calcium (Ca)	1.19	16.62
Magnesium (Mg)	3.72	4.54
Sodium (Na)	30.53	6.98
Potassium (K)	1.11	2.55
Bicarbonate (HCO_3)	0.42	31.90
Sulfate (SO_4)	7.67	12.41
Chloride (Cl)	55.16	8.64
Nitrate (NO_3)	—	1.11
Bromide (Br)	0.20	—
TOTAL	100.00	100.00

Where do the Ions go?

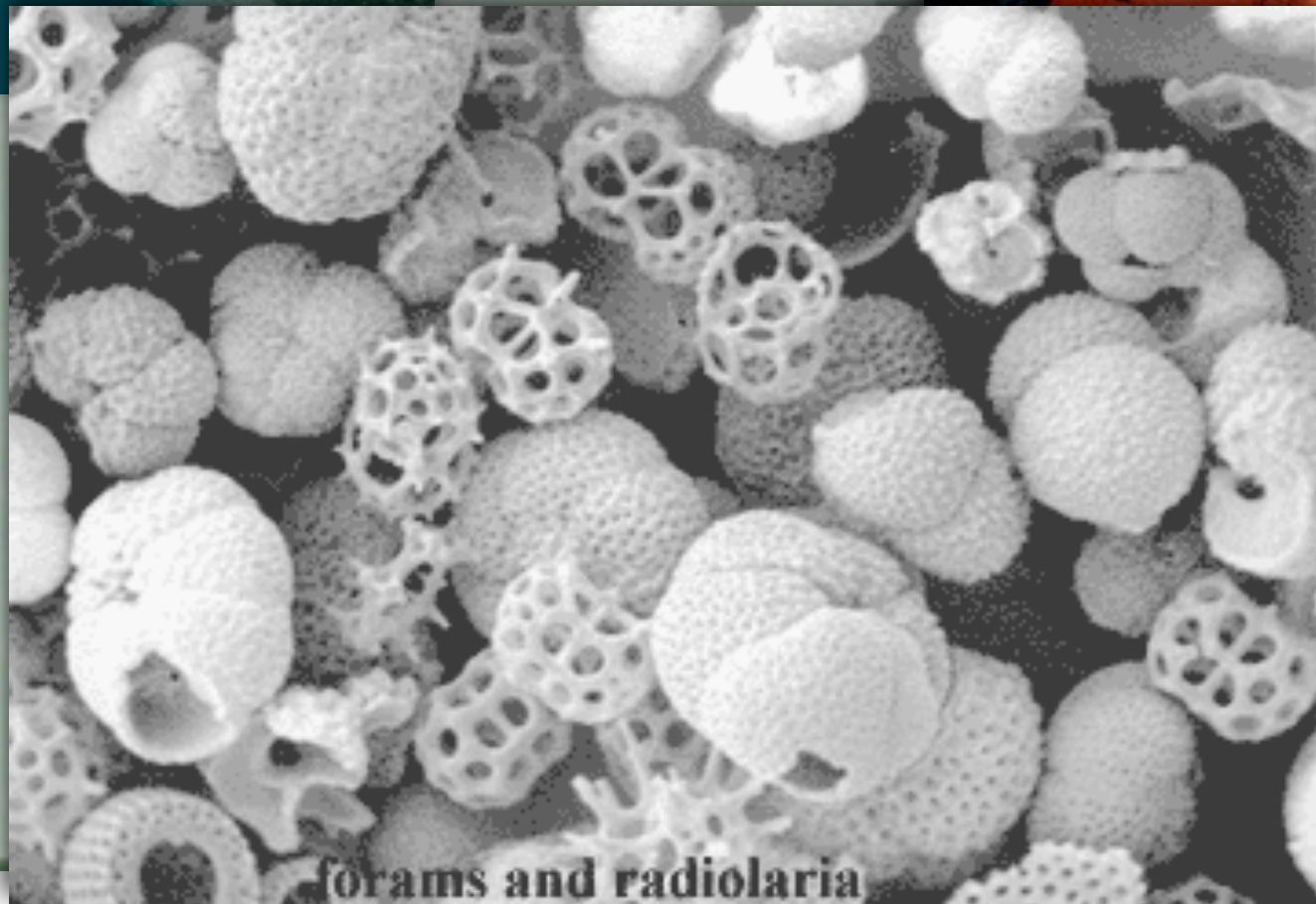
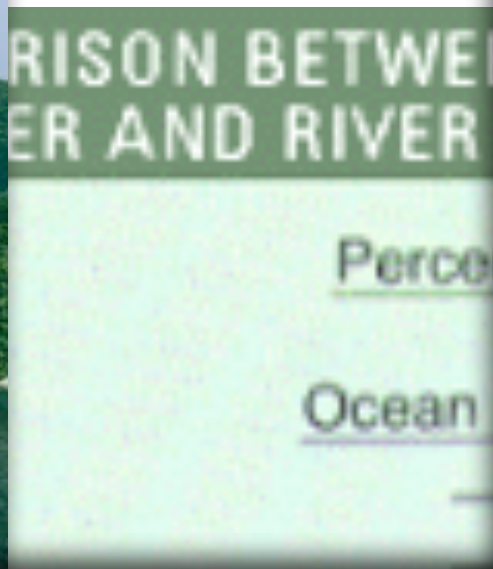
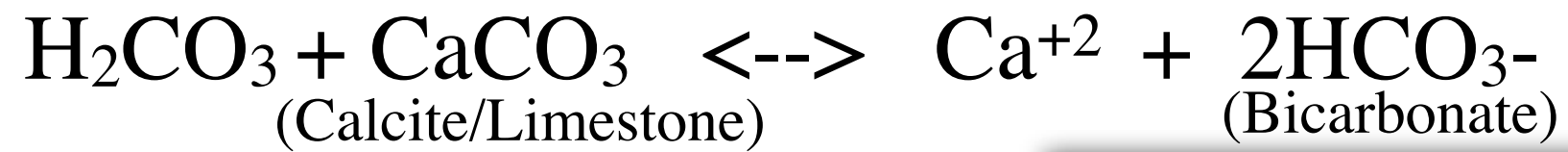


COMPARISON BETWEEN
OCEAN AND RIVER



	Percent Ocean	
	1.1	
Magnesium (Mg)	3.72	4.54
Sodium (Na)	30.53	6.98
Potassium (K)	1.11	2.55
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TOTAL	100.00	100.00

Where do the Ions go?



forams and radiolaria

Where do the ions go?

Chemical Sedimentary Rocks

Limestone (CaCO_3)

Chert (SiO_2)

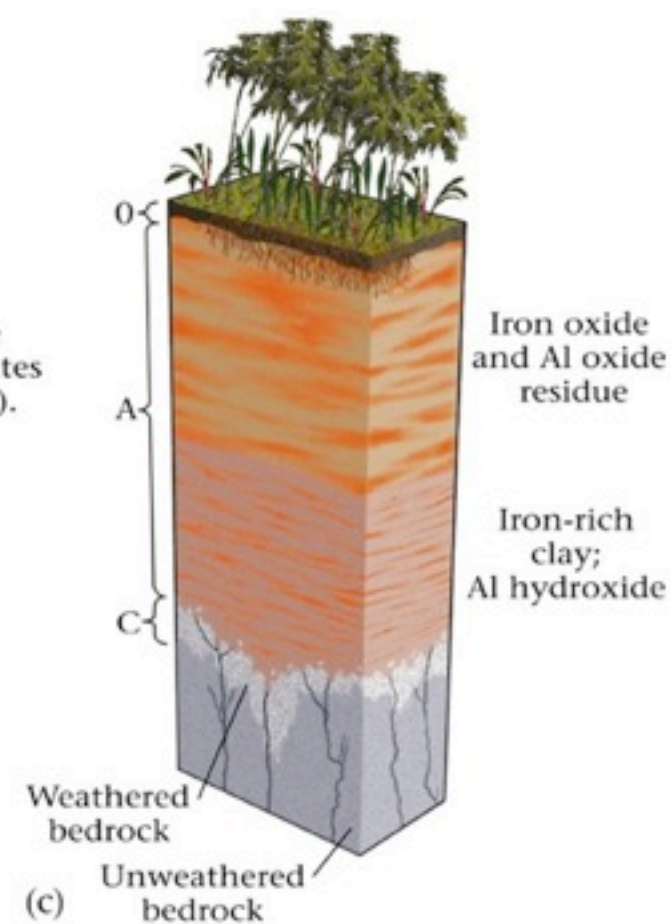
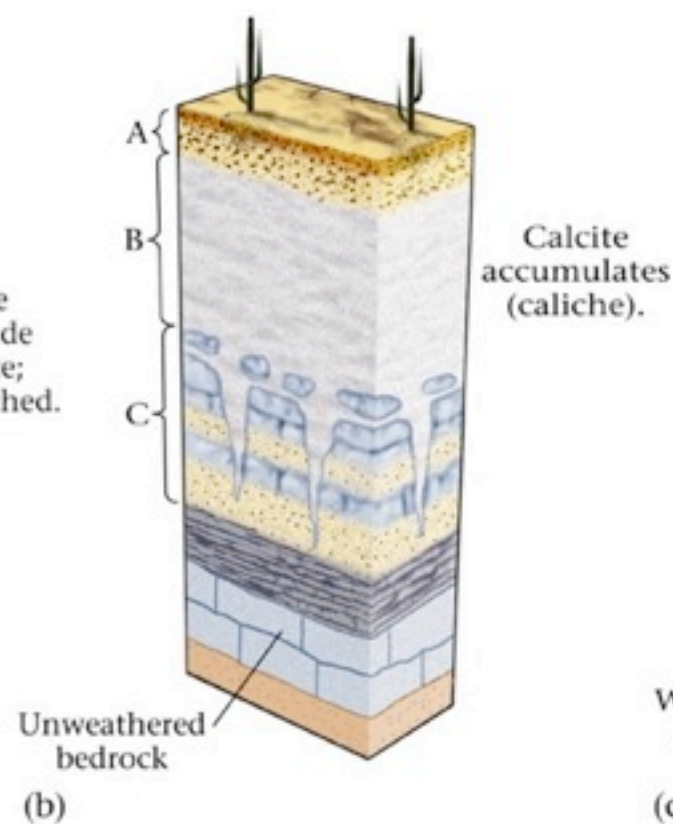
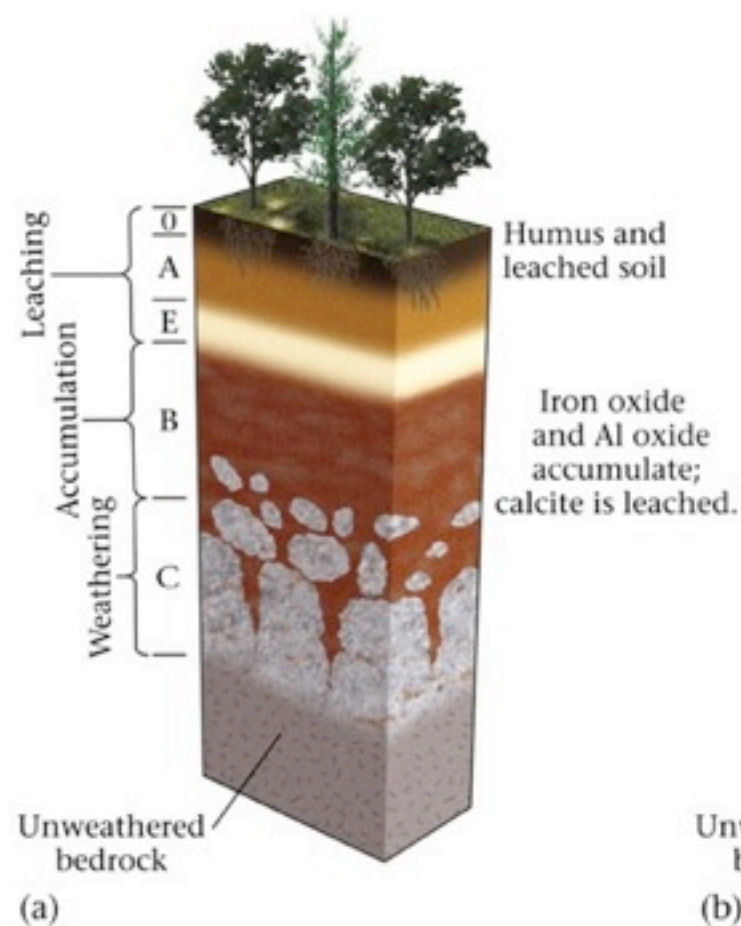
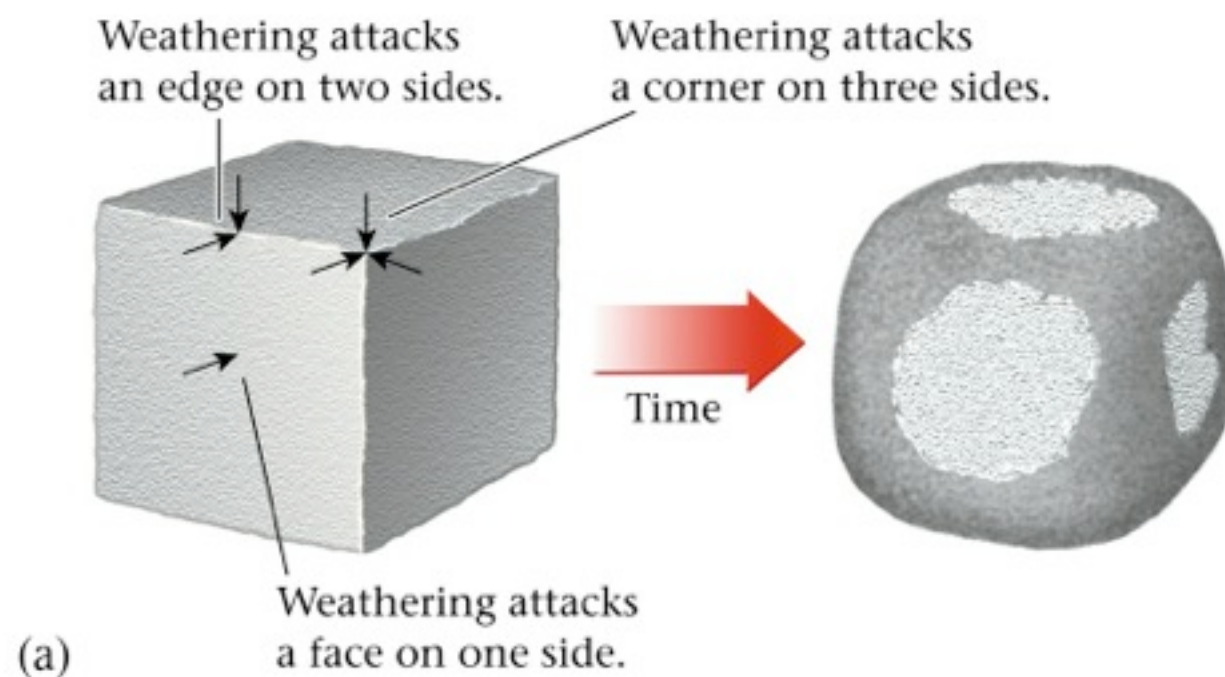
Cements

Calcite (CaCO_3)

Silica (SiO_2)

Hematite (Fe_2O_3)

Subsurface Weathering

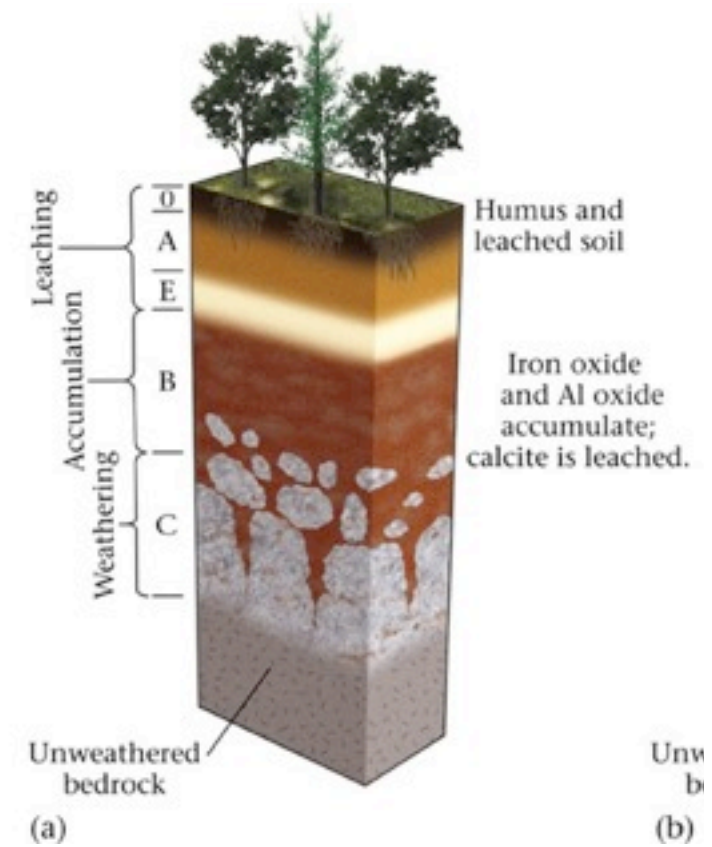


Weathering of Common Rocks

Rock	Primary Minerals	Residual Minerals*	Leached Ions
Granite	Feldspars	Clay Minerals	Na^+, K^+
	Micas	Clay Minerals	K^+
	Quartz	Quartz	---
	Fe-Mg Minerals	Clay Minerals + Hematite + Goethite	Mg^{+2}
Basalt	Feldspars	Clay Minerals	$\text{Na}^+, \text{Ca}^{+2}$
	Fe-Mg Minerals	Clay Minerals	Mg^{+2}
	Magnetite	Hematite, Goethite	---
Limestone	Calcite	None	$\text{Ca}^{+2}, \text{CO}_3^{-2}$

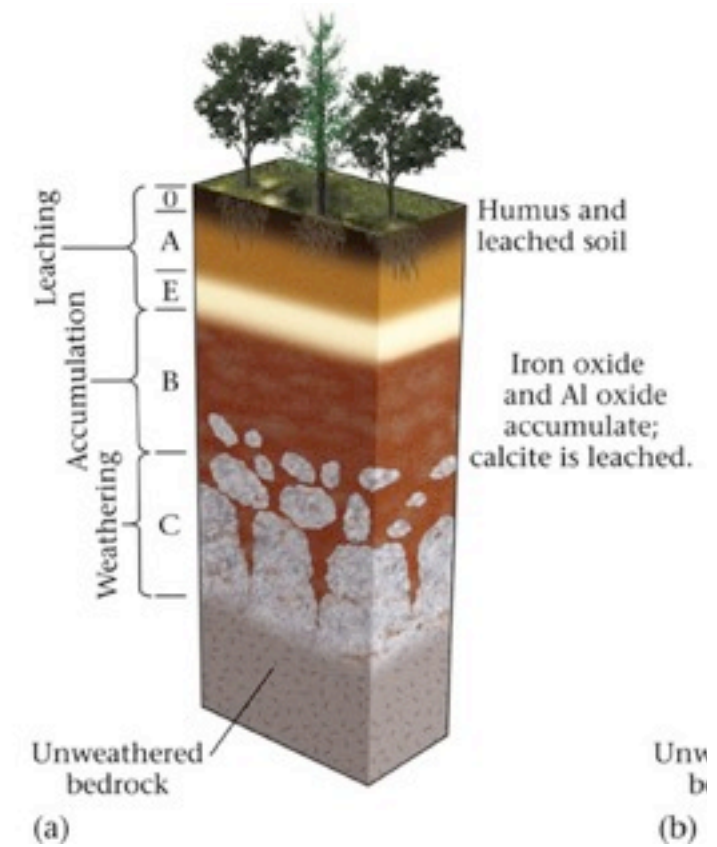
Subsurface Weathering

Chemical Weathering



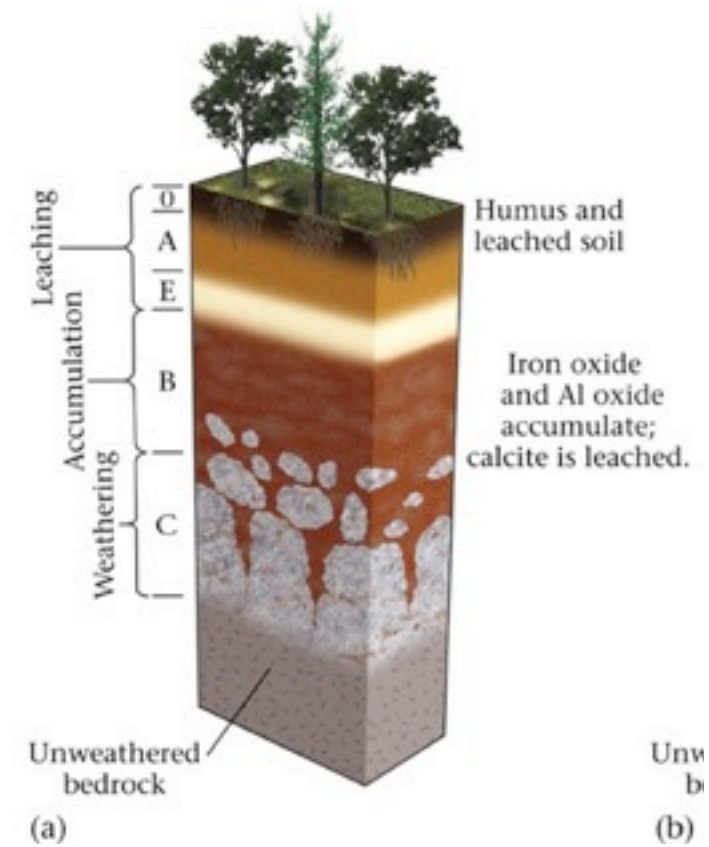
Subsurface Weathering

Chemical Weathering



Subsurface Weathering

Chemical Weathering

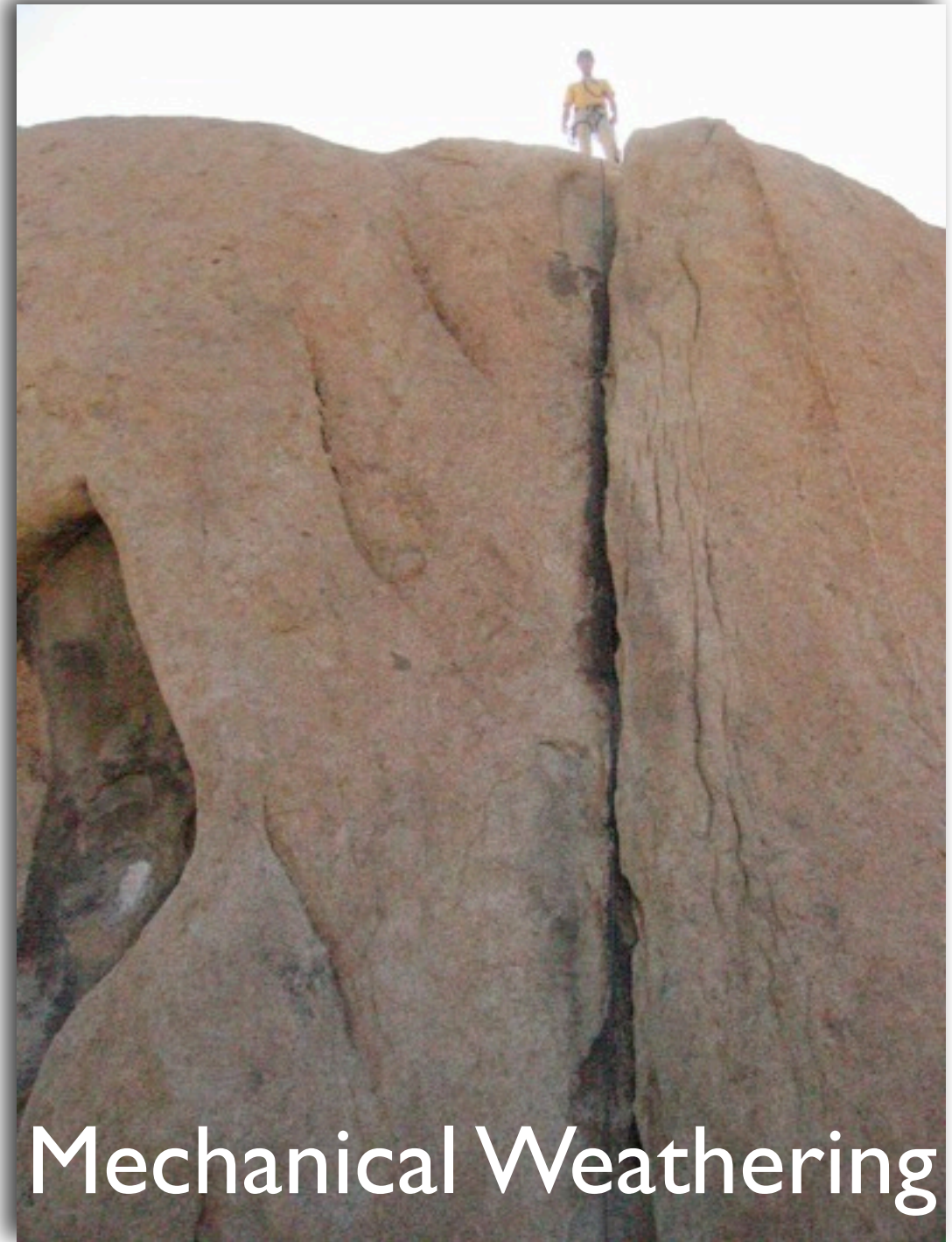


Enjoying Weathering

Spelunking Carlsbad Caverns.



Dissolution



Mechanical Weathering

Cool crack climbs,
Joshua Tree NP

Erosion

Refers to the combination of processes that separates weathered bedrock (sediment) is transported to a new location.



River Erosion

Glacial Erosion



Surface Run off



Coastal erosion



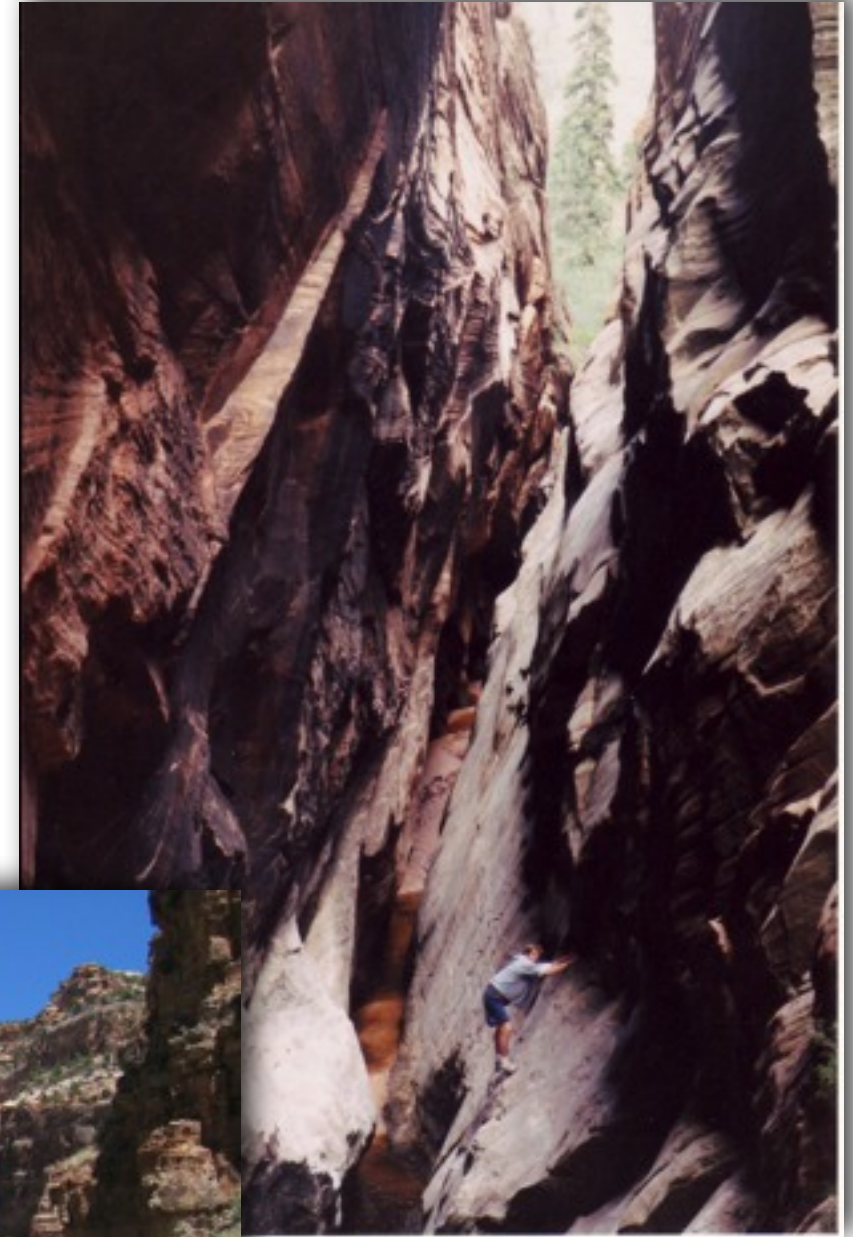
Wind Erosion

Enjoying Erosion

Canyoneering, Zion NP



Hanging around
Grand Canyon



Yampa River, Utah

